SHRI DAVARA UNIVERSITY

NAYA RAIPUR (C.G.)

PROGRAMME- CURRICULUM

FOR

BECHELOR IN SCIENCES

(Chemistry, Physics, Zoology (PCB)

SEMESTER-I

AS PER NEW EDUCATION POLICY-2020

AND

NATIONAL EDUCATION POLICY-2025

FOUR YEAR UNDERGRADUATE PROGRAMME-2024-25

(EFFECTIVE FROM THE SESSION-2024-2025)

SHRI DAVARA UNIVERSITY NAYA RAIPUR

INTRODUCTION OF THE DEPARTMENT: -

Department of Chemistry

Introduction

The Department of Chemistry is a vibrant community of scholars, researchers, and students dedicated to advancing our understanding of the chemical sciences. Our department offers undergraduate and postgraduate programs in chemistry, providing students with a comprehensive education in the principles and applications of chemistry.

Mission

Our mission is to provide students with a rigorous and well-rounded education in chemistry, preparing them for careers in research, industry, education, and beyond. We strive to create an inclusive and supportive environment that encourages collaboration, creativity, and intellectual curiosity.

Research Areas

Faculty and students in the Department of Chemistry engage in cutting-edge research in various areas, including:

1. Organic Chemistry: Synthesis and characterization of organic compounds.
2. Inorganic Chemistry: Study of inorganic compounds and their applications.
3. Physical Chemistry: Investigation of the physical

principles underlying chemical phenomena.

4. Analytical Chemistry: Development and application of analytical techniques.

Department of Physics

Introduction

The Department of Physics is a dynamic community of Mechanics physiologists, researchers, and students dedicated to exploring the fascinating world of physics. Our department offers undergraduate and postgraduate programs in physics, providing students with a comprehensive education in physics.

Mission

Our mission is to provide students with a rigorous and well-rounded education in physics, preparing them for careers in research, conservation, education, and beyond. We strive to create an inclusive and supportive environment that encourages collaboration, creativity, and intellectual curiosity.

Research Areas

Faculty and students in the Department of Physics engage in cutting-edge research in various areas, including:

1. Physics Systematics: Study of physics classification, evolution, and diversity.

2. Physics Mechanism : Investigation of growth,

development, and responses to physics.

Department of Zoology

Introduction

The Department of Zoology is a vibrant community of animal biologists, researchers, and students dedicated to exploring the fascinating world of animals. Our department offers undergraduate and postgraduate programs in zoology, providing students with a comprehensive education in animal biology.

Mission

Our mission is to provide students with a rigorous and well-rounded education in zoology, preparing them for careers in research, conservation, education, and beyond. We strive to create an inclusive and supportive environment that encourages collaboration, creativity, and intellectual curiosity.

Research Areas

Faculty and students in the Department of Zoology engage in cutting-edge research in various areas, including:

1. Animal Systematics: Study of animal classification, evolution, and diversity.
2. Animal Physiology: Investigation of animal growth, development, and responses to environmental stimuli.
3. Animal Ecology: Study of animal interactions with their environment and other organisms.
4. Animal Behavior: Study of animal behavior,

including social behavior, communication, and learning.

VISION OF DEPARTMENT: -

Department of Chemistry

Vision

To be a leading department of chemistry, recognized for its academic excellence, innovative research, and commitment to fostering a community of scholars who can address the complex chemical challenges of the 21st century.

Objectives

1. To provide students with a world-class education in chemistry, emphasizing both theoretical foundations and practical applications.
2. To conduct cutting-edge research in chemistry, focusing on areas of national and global importance.
3. To foster collaborations with other departments, institutions, and industries to promote interdisciplinary research and innovation.
4. To develop and maintain state-of-the-art research facilities and instrumentation.

Department of Physics

Vision

To be a premier department of physics, dedicated to advancing our understanding of physics and addressing

the critical challenges facing our mechanism, including deferments physics branches.

Objectives

1. To provide students with a comprehensive education in physics, emphasizing both theoretical foundations and practical applications.

2. To conduct innovative research in physics, focusing on areas of national and global importance.

3. To develop and maintain a diverse collection of physical data.

4. To foster collaborations with other departments, institutions, and industries to promote interdisciplinary research and innovation.

Department of Zoology

Vision

To be a leading department of zoology, recognized for its academic excellence, innovative research, and commitment to fostering a community of scholars who can address the complex challenges facing animal populations and ecosystems.

Objectives

1. To provide students with a world-class education in zoology, emphasizing both theoretical foundations and practical applications.

2. To conduct cutting-edge research in animal biology, focusing on areas of national and global importance.

3. To develop and maintain state-of-the-art research facilities and instrumentation.

4. To foster collaborations with other departments, institutions, and industries to promote interdisciplinary research and innovation.

SCOPE OF DEPARTMENT: -

Department of Chemistry

Scope

1. Research and Development: Opportunities exist in various fields like medicinal chemistry, materials science, environmental chemistry, and analytical chemistry.

2. Industry and Manufacturing: Chemistry graduates can work in various industries like pharmaceuticals, petrochemicals, and materials manufacturing.

3. Environmental Conservation: Chemists can work in environmental monitoring, conservation, and sustainability.

4. Education and Academia: Chemistry graduates can pursue teaching and research careers in academic institutions.

5. Government and Policy: Chemists can work in government agencies, regulatory bodies, and policy-making institutions.

Department of Physics

Scope

1. Research and Development: Opportunities exist in

various fields like physics.

2. Education and Academia: Physics graduates can pursue teaching and research careers in academic institutions.

5. Government and Policy: Physics can work in government agencies, regulatory bodies, and policy-making institutions.

Department of Zoology

Scope

1. Research and Development: Opportunities exist in various fields like animal physiology, ecology, evolution, and conservation biology.

2. Wildlife Conservation and Management: Zoology graduates can work in wildlife conservation, management, and research institutions.

3. Animal Health and Veterinary Science: Zoologists can work in animal health, veterinary science, and animal biotechnology industries.

4. Education and Academia: Zoology graduates can pursue teaching and research careers in academic institutions.

5. Government and Policy: Zoologists can work in government agencies, regulatory bodies, and policy-making institutions.

PROGRAMME OUTCOME: -

1**. Knowledge and Understanding:** Demonstrate advanced knowledge and understanding of scientific principles, theories, and concepts in their chosen field.

2. **Critical Thinking and Analysis:** Apply critical thinking and analytical skills to evaluate scientific data, theories, and methodologies.

3. **Research and Problem-Solving:** Design, conduct, and analyze research experiments, and apply scientific principles to solve complex problems.

4. **Communication and Collaboration:** Communicate complex scientific ideas effectively to various audiences, and collaborate with others in a research or professional setting.

5. **Scientific Literacy and Critical Evaluation:** Evaluate the scientific literature, identify gaps in knowledge, and propose new research directions.

6. **Professional Development and Leadership:** Demonstrate leadership skills, manage projects, and develop a professional network in their chosen field.

7. **Knowledge and Understanding:** Demonstrate a solid understanding of scientific principles, theories, and concepts in their chosen field.

8. **Critical Thinking and Analysis:** Apply critical thinking and analytical skills to evaluate scientific data and theories.

9. **Scientific Literacy and Communication:** Communicate scientific ideas effectively to various audiences, and demonstrate an understanding of the scientific method.

10. **Problem-Solving and Laboratory Skills:** Apply scientific principles to solve problems, and demonstrate laboratory skills and safety protocols.

11. **Teamwork and Collaboration:** Collaborate with others in a laboratory or project setting, and demonstrate

an understanding of the importance of teamwork in science.

12. **Preparation for Further Study or Employment:** Demonstrate preparation for further study or employment in a scientific field, and exhibit a commitment to lifelong learning.

COURSE OUTCOME: -

Department of Chemistry

Course Outcomes

1**. Knowledge of Chemical Principles:** Understand and apply fundamental chemical principles, theories, and concepts.

2. **Laboratory Skills:** Develop laboratory skills, including experimentation, data analysis, and safety protocols.

3. **Analytical and Problem-Solving Skills:** Apply analytical and problem-solving skills to solve chemical problems and analyze data.

4. **Communication Skills:** Communicate chemical concepts and research findings effectively through written and oral presentations.

5. Critical Thinking and Scientific Literacy: Evaluate scientific literature, identify gaps in knowledge, and propose new research directions.

Department of Physics

Course Outcomes

1. **Knowledge of Physical science:** Understand and

apply fundamental principles of physics, including deferments branche.

2. **Laboratory and Field Skills:** Develop laboratory and field skills, including plant collection, preservation, and experimentation.

4. Ecological and Environmental Awareness: Understand the importance of physics in ecosystems and the impact of human activities on physical science.

5. **Communication and Critical Thinking:** Communicate physical concepts and research findings effectively and critically evaluate scientific literature.

Department of Zoology

Course Outcomes

1. **Knowledge of Animal Biology:** Understand and apply fundamental principles of animal biology, including morphology, anatomy, physiology, and ecology.

2. **Animal Identification and Classification:** Identify and classify animals using morphological, anatomical, and molecular characteristics.

3. **Laboratory and Field Skills:** Develop laboratory and field skills, including animal collection, preservation, and experimentation.

4. **Ecological and Environmental Awareness:** Understand the importance of animals in ecosystems and the impact of human activities on animal diversity and ecology.

5. **Communication and Critical Thinking:** Communicate zoological concepts and research findings

effectively and critically evaluate scientific literature.

UNIT OUTCOME: -

Department of Chemistry

Unit 1: Atomic Structure and Chemical Bonding

1. Explain the structure of atoms and molecules.

2. Describe the types of chemical bonds and their properties.

3. Apply knowledge of atomic structure and chemical bonding to predict chemical behavior.

Unit 2: Thermodynamics and Kinetics

1. Understand the laws of thermodynamics and their applications.

2. Explain the principles of chemical kinetics and reaction mechanisms.

3. Apply thermodynamic and kinetic principles to solve chemical problems.

Unit 3: Organic Chemistry

1. Identify and classify organic compounds.

2. Explain the mechanisms of organic reactions.

3. Apply knowledge of organic chemistry to synthesize and analyze organic compounds.

Unit 4: Analytical Chemistry

1. Understand the principles of analytical chemistry

techniques.

2. Apply analytical techniques to analyze and identify chemical substances.

3. Interpret analytical data to solve chemical problems.

Department of Physics

Mechanics

1. Kinematics: Students will be able to describe the motion of objects in terms of position, velocity, and acceleration.

2. Dynamics: Students will be able to apply Newton's laws of motion to solve problems involving force, mass, and acceleration.

3. Energy and Momentum: Students will be able to analyze the conservation of energy and momentum in physical systems.

Electromagnetism

1. Electric Fields: Students will be able to calculate the electric field due to a point charge and a distribution of charges.

2. Magnetic Fields: Students will be able to calculate the magnetic field due to a current-carrying wire and a magnetic dipole.

3. Electromagnetic Waves: Students will be able to describe the properties of electromagnetic waves and their applications.

Thermodynamics

1. Temperature and Heat: Students will be able to define temperature and heat, and describe the laws of thermodynamics.

2. Thermodynamic Systems: Students will be able to analyze thermodynamic systems, including ideal gases and heat engines.

3. Entropy and Free Energy: Students will be able to calculate entropy and free energy changes in thermodynamic systems.

Quantum Mechanics

1. Wave-Particle Duality: Students will be able to describe the wave-particle duality of matter and energy.

2. Schrödinger Equation: Students will be able to solve the Schrödinger equation for simple systems, such as the particle in a box.

3. Quantum Systems: Students will be able to analyze quantum systems, including the hydrogen atom and quantum harmonic oscillator.

Statistical Mechanics

1. Probability and Statistics: Students will be able to apply probability and statistical concepts to physical systems.

2. Thermodynamic Systems: Students will be able to analyze thermodynamic systems using statistical mechanics.

3. Phase Transitions: Students will be able to describe

phase transitions and critical phenomena using statistical mechanics.

Computational Physics

1. Numerical Methods: Students will be able to apply numerical methods, such as finite difference and Monte Carlo methods, to solve physical problems.

2. Programming Languages: Students will be able to write programs in languages such as Python, C++, and MATLAB to solve physical problems.

3. Data Analysis: Students will be able to analyze and visualize data using computational tools.

Experimental Physics

1. Experimental Design: Students will be able to design and conduct experiments to test physical hypotheses.

2. Data Analysis: Students will be able to analyze and interpret data from experiments.

3. Instrumentation: Students will be able to use and maintain laboratory instrumentation, such as oscilloscopes and spectrometers.

Department of Zoology

Unit 1: Animal Morphology and Anatomy

1. Identify and describe animal morphological and anatomical features.

2. Explain the functions of animal tissues and organs.

3. Apply knowledge of animal morphology and anatomy to understand animal development and evolution.

Unit 2: Animal Physiology

1. Understand the principles of animal physiology, including nervous and circulatory systems.

2. Explain the mechanisms of animal growth and development.

3. Apply knowledge of animal physiology to solve problems in animal health and welfare.

Unit 3: Animal Ecology

1. Understand the principles of animal ecology, including population dynamics and community structure.

2. Explain the interactions between animals and their environment.

3. Apply knowledge of animal ecology to understand and manage ecosystems.

Unit 4: Animal Genetics and Evolution

1. Understand the principles of animal genetics and evolution.

2. Explain the mechanisms of animal genetic variation and evolution.

3. Apply knowledge of animal genetics and evolution to understand animal diversity and adaptation.

SHRI DAVARA UNIVERSITY NAYA RAIPUR

SEMESTER I

S.NO	COURSE CODE	COURSE TITLE	TEACHING HOURS PER WEEK				EXAMINATION SCHEME				
DISCIPLINE SPECIFIC COURSE (DSC)			L	T	P	C	THEORY		PRACTICAL		TOTAL MARKS
							EX	IN	EX	IN	
1.	CHSC-01T	Fundamental Chemistry-I	2	1	0	3	70	30	-	-	100
2.	PHSC-O1T	Mechanics	2	1	0	3	70	30	-	-	100
3.	ZOSC-01T	Life on Earth and Unique Attributes of Animal Kingdom	2	1	0	3	70	30	-	-	100
GENERAL ELECTIVE (GE)											
4.	SCGE-10	Introduction to Political Theory	3	1	0	4	70	30	-	-	100
ABILITY ENHANCEMENT COURSE (AEC)											
5.	AEC-01	Communicative of English and Soft Skills	2	0	0	2	35	15	-	-	50
VALUE ADDITION COURSE (VAC)											
6.	VAC-01T	Computer fundamental & MS Office	1	1	0	2	35	15	-	-	50
PRACTICAL (LAB)											
7.	CHSC-01P	Fundamental Chemistry-I LAB	0	0	2	1	-	-	35	15	50
8.	PHSC-O1P	Mechanics-LAB	0	0	2	1	-	-	35	15	50
9.	ZOSC-0P	Life on Earth and Unique Attributes of Animal Kingdom LAB	0	0	2	1	-	-	35	15	50
Total Contact hours Per Week:30		Total credit:				20	Total mark				650

SHRI DAVARA UNIVERSITY NAYA RAIPUR

FOUR YEAR UNDERGRADUATE PROGRAM (2024-28)

DEPARTMENT OF CHEMISTRY

COURSE CURRICULUM

PART-A: Introduction			
Programme: Bachelor in Sciences (Certificate/Diploma/Degree Honors)		**Semester-1**	**Session: 2024-2025**
Course Code	**CHSC-01T**		
Course Title	**FUNDAMENTAL CHEMISTRY-I**		
Course Type	**Discipline Specific course (DSC)**		
Pre-requisite(if any)	**As per program**		
Course Learning. Outcomes (CLO)	**At the end of this course, the students will be able to** ➢ **The know the contributions of ancient Indian scientists, study atomic , and periodic properties.** ➢ **To explore the concept of chemical bonding, including ionic and covalent bonding, hybridization, molecular orbital theory and intermolecular interaction.** ➢ **To learn about reaction mechanisms of inorganic reactions and their stoichiometry's** ➢ **To understand basics principles of organic chemistry**		
Credit Value	**3 Credits**	**Credit =15 Hours-learning & Observation**	
Total Marks	**Max. Marks:=100**	**Min Passing Marks: 40**	

PART -B: Content of the Course		
Total No. of Teaching-learning Periods (01 Hr. per period) -45 Periods (45 Hours)		
Unit	Topics (Course contents)	
I	**A. Chemistry in Ancient India**: (a) Chemical techniques in ancient India: General Introduction (b) Contribution of ancient Indian scientists in chemistry, e.g, metallurgy, byes, pigments, cosmetics, Ayurveda, Charak Sanhita . **Ancient Indian Chemist-** Their Contribution and Books- Rishi Kanad, Aacharya Nagarjuna, Vagbhatta, Govindacharya. Yashodhar, Ramchandra, Somadava, Gopalbhatta Indian Chemist of 19th century-Aacharya Prafulla Chandra Ray-His Contribution and work for Indian Chemistry. **B. Atomic Structure and Periodic Properties**: (i) Review of Bohr's theory and us Sumitations. Dual nature of particles and waves, de Broglie's equation, Heisenberg's Uncertainty principle and its significance. (ii) Quantum numbers and their significance Rules for Tilling electrons in various orbital's, Pauli's Exclusion Principle, Hund's rule of maximum multiplicity, Aufbau principle and its limitations. Electronic configurations of the atoms Stability of half-filled and completely filled orbitals, concept of exchange energy Relative energies of atomic orbitals. Anomalous electronic configurations flective nuclear charge (UNC), shielding or screening effect. Slater rules, Atomic and lonic radii ionization energy and factors	12

	affecting ionization energy. Electron affinity. Electronegativity Pauling/Mulliken's electro negativity scales Relation of lector negativity with hybridization.	
II	**Chemical Bonding**-1A**) Ionie Bonding:** General characteristics of ionic bondong Sonic **Bonding & Energy:** Lattice and solvation energies and their importance in the pantear context of stability and solubility of ionic compounds. **Bors-Haber Cycle and its Applications**: Covalent character in ionic compounds patlarizing power and polarizability, Fajan's moles. Covalent Beading: Lewis structures, Valence Bond theory, Hybridization (Concept and types with suitable examples), dipole moment and parentage tonic character .Value shell electron pair repulsion theory (VSEPR) and structure of NH_3,H_2O, SF_4,CIF_3,PCl_5,SF_6,XeF_2,XeF_6,XeO_3,$XeOF_4$, XeF_4. **Chemical Bonding-II** **A) MO theory:** LCAO method-criteria of orbital overlapping, types of molecular orbital's-σ ,π and, δ-MOs, formation of σ - and π -MOs and their, schematic illustration qualitative MO energy level diagram of homo- (N; & O:(including peroxide, superoxide) and hetero-diatomic molecules (NO, CO), magnetic properties, bond order and stability of molecules and ions. **B)** Weak Chemical Forces: van der Waals forces, ion-dipole forces, dipole-dipole interactions, ion-induced dipole interactions, dipole-induced dipole interactions. Repulsive forces, Hydrogen bonding (theories of hydrogen bonding, valence bond treatment).	11
III	**A. Chemical properties of s-block metals** Reaction with water, air, and nitrogen, Anomalous behavior of Li and Be, Compounds of -block metals: Oxides, hydroxides, peroxides, and super oxides (preparation and properties) Complexes of s-block metals, Complexes with crown ethers **B. Chemistry of p-Block Elements** **Boron group:** Hydrides (classification of beranes), Diborane (preparation, properties, and structure elucidation), Borazine (preparation and structure) **Carben group:** Carbides (salt-like carbides, interstitial carbides, covalent carbides), Silicates (classification, three-dimensional silicates - properties and structures) **Nitrogen group:** Hydrides of Nitrogen (hydrazine, hydroxylamine, hydrazoic acid) Structure of oxides of nitrogen (N,O, NO, NO2, N2O4, and N_2Os), Structure of oxyacids of nitrogen (HNO3, HNOJ, H_2NO1,), Nitrides (classification, preparation, properties, and uses) Structure of Oxides and oxoacids of phosphorus: (P_2O_3, P_2O) H_2PO, H_2POS, H_3PO HPO **Halogen:** Hydrides, Oxides and oxyacids of halogens (structure only) - Inter halogen compounds and pseudo halogen.	11
IV	**A. Chemistry of Noble Gases:** Chemical properties of the noble gases, chemistry of xenon, Structure bonding an xenon compounds. **B. Theoretical Principles in Qualitative Analysis (H_2S Scheme):** Basis principles involves in the analysis of cat ions and anions and solubility products, common ion effect. Principles involves in separation of cat ions into groups and choice of group reagents. Interfering anions (fluoride, borate, oxalate and phosphate) and need to remove them after Group II.	11
Keywords	Ancient Indian Chemistry , Atomic Structure, Periodic properties, Chemical Bonding , S - & - P block elements , Chemistry of Noble Gases. Theoretical Principles in Qualitative Analysis	
Signature of Convener & Members (CBoS)		

SHRI DAVARA UNIVERSITY NAYA RAIPUR

FOUR YEAR UNDERGRADUATE PROGRAM (2024-28)

DEPARTMENT OF CHEMISTRY

COURSE CURRICULUM

PART-C: Learning Resources

Text Books, Reference Books and Others

Text Books Recommended-

1. Puri, B. R., Sharma, L. R., & Kalis, K. C. (2018) Principles of Inorgante Chemistry. Nagin Chand and Co., New Delhi.
2. Satyaprakash, G., Tali, S. K., Basu, S. K., & Maden, R. D. (2017). Advanced Inorganic Chemistry (Vol. 1, 5th Ed.). S. Chand & Company.
3. 3. Lee, J. D. (2010). Concise Inorganic Chemistry (5th Ed.). Blackwell Science.
4. Housecroft, C. E., & Sharpe, A. G. (2012). Inorganic Chemistry (4th Ed.). Pearson Education Limited.
5. 5. Ray, Acharya Prafulla Charndra, History of Chemistry in Ancient And Medieval India, Chowkhamba Krishnadas Academy (Reprint 2004).

Reference Books Recommended-

1. Cotton, F. A., Wilkinson, G., & Gaus, P. L. (2002). Basic Inorganic Chemistry (3rd Ed.). John Wiley & Sons.
2. Douglas, B. E., Mcdaniel, D. T., & Alexander, J. J. (1994). Concepts and Models Of Inorganic Chemistry (3rd Ed.). John Wiley & Sons.
3. Huheey, J. E., Keiter, E. A., & Keiter, R. L. (1993). Inorganic Chemistry (4th Ed.). Harpercollins.
4. College Publishers. 4. Shriver, D. F., Atkins, P. W., & Langford, C. H. (2010). Inorganic Chemistry (5th Ed.). W. H. Freeman And Company.
5. Moeller, T. (1990). Inorganic Chemistry: A Modern Introduction. Wiley.

Online Resources-

- e-books and e-learning portals
- https://bit.lv/3AvV3mZ
- https://bit.ly/30V85z
- https://bit.ly/3C9PXPS
- https://bit.ly/301p9rZ
- https://bit.ly/BPnwqe

Online Resources-
e-sources/e-books and e-learning portals

ART -D: Assessment and Evaluation

Suggested Continuous Evaluation Methods:
Maximum Marks: 100 Marks
Continuous Internal Assessment (CIA): 30 Marks
End Semester Exam (ESE): 70 Marks

Continuous Internal Assessment (CIA): 30 (By Course Teacher)	Internal Test/Quiz:20+20 Assignment/ Semenar-10 Total Marks-30	Better marks out of the two Tot Quiz + obtained marks in Assignment shall be considered against 15 Marks

End Semester Exam (ESE):70	Two section A&B Section A :Q1 Objective 10∗1=10 Marks Q2 Short answer type-5*4=20 Section B : Descriptive answer type qts 1 out of 2frm each- 4*10=40 Marks

Signature of Convener & Members (CBoS)

SHRI DAVARA UNIVERSITY NAYA RAIPUR

FOUR YEAR UNDERGRADUATE PROGRAM (2024-28)

DEPARTMENT OF CHEMISTRY

COURSE CURRICULUM

PART- A: Introduction			
Program: Bachelor in Sciences **(Certificate/Diploma/Degree/Honors**		**Semester-1**	**Session: 2024-2025**
Course Code	**CHSC-01P**		
Course Title	**Lab. Course -01 (FUNDAMENTAL CHEMISTRY)**		
Course Type	**Laboratory course**		
Pre-requisite(if any)	**As per program**		
Course Learning. Outcomes (CLO)	**At the end of this course, the students will be able to** ➢ **Analyze mixtures for catlons (NH, Pb, etc.) & anions (CO, S, etc.) using H3S or other methods.** ➢ **Perform ürimetric analysis (standardization, unknown conc determination).** ➢ **stimate the concentration of acetic acid in vinegar (using NaOH), alkali content in soaps/detergents.** ➢ **Utilize complexometric titrations for calcium (Ca"), water hardness, Fe/Fe", and Cu.**		
Credits Value	**1 Credits**	**Credit =30 Hours Laboratory or Field learning/Training**	
Total Marks	**Max. Marks:50**	**Min Passing Marks: 20**	

PART-B: Content of the Course		
Total No. of learning-Training/performance Periods: 30 Periods (30 Hours)		
Module	Topics (Course contents)	No. of Period
Lab/ field Training/ Experiment Content of Course	**QUALITATIVE INORGANIC MIXTURE ANALYSIS**: Inorganic mixture Training analysis containing up to four lonic species (two cations and two anions) using are interfering and Experiment CHS (hydrogen sulfide) or other appropriate methods (Excluded of Course insoluble salts) Cations and anions that may be encountered include: **Cations:** NH_4^+, Pb^{2+}, Bi, Cu^+ Cd, Fe/Fe, Al, Co, Ni, Mn, Zn, Ba S, Ca, Na **Anines:** CO, S, 50, NO, CILCOO, CT, Br, I, NO, 50 (Spot tesis may be used wherever feasible.) TIERIMETRIC ANALYSIS Standardize sodium hydroxide solution using a standard oxalic acid solution Determine the concentration of hydrochloric acid (HCI) solution using standardized sodium hydroxide solution as an intermediate	30
Keywords	Qualitative Analysis, Titrimetric Analysis.	
Signature of Convener & Members (CBoS)		

SHRI DAVARA UNIVERSITY NAYA RAIPUR

FOUR YEAR UNDERGRADUATE PROGRAM (2024-28)

DEPARTMENT OF CHEMISTRY

COURSE CURRICULUM

PART-C: Learning Resources

Text Books, Reference Books and Others

Text Books Recommended-

1. Gurtu, J. N., & Kapoor, R. (1987). Experimental Chemistry. S. Chand & Co
2. Bajpai, D. N., Pandey, O. P., & Giri, S. (2013). Practical Chemistry. S. Chand & Co.
3. 3. Ahluwalia, V. K., Dhingra, S., & Dhingram, S. (2005). College Practical Chemistry. Universities.
4. Kamboj, P. C. (2014). Advanced University Practical Chemistry (Part 1). Vishal Publishing Co. 5. Fultariya, C., & Harsora, J. (2017). Volumetric Analysis: Conceptt and Experiments.

Reference Books Recommended-

1. Mcpherson, P. A. (2015). Practical Volumetric Analysis. Royal Society Of Chemistry.
2. Shobha, R., & Banani, M. (2017). Essentials of Analytical Chemistry. Pearson.
3. Venkateswaran, V., Veeraswamy, R., & Kulandaivelu, A. R. (2004). Basic Principles Of Practical Chemistry (2nd Ed.). S. Chand Publications.
4. Sundaram, S., & Raghavan, K. (1996), Practical Chemistry. S. Viswanathan Co. Pvt. 5. Svehla, G. (2011). Vogel's Textbook of Inorganic Qualitative Analysis (7th Ed.). Pearson Education

Online Resources-

- E-resources/e-books and e-learning portals
- http://www.swayam.ac.in
- http://www.ignou.ac.in
- www.egyankosh.ac.in
- www.litm.ac.in
- www.eskillindia.org
- www.eshiksha.mp.gov.in
- www.vlab.co.in

Online Resources-

e-sources/e-books and e-learning portals

- https://www.ncbi.nlm.nih.gov/pmc/articles/PMC5871155/
- https://cms.botany.org/home/careers-jobs/careers-in-botany/arcas-of-specialization- in-botany.html

PART -D: Assessment and Evaluation

Suggested Continuous Evaluation Methods:
Maximum Marks: 50 Marks
Continuous Internal Assessment (CIA): 15 Marks
End Semester Exam (ESE): 35 Marks

Continuous Internal Assessment (CIA): 15 (By Course Teacher)	Internal Test/Quiz:10+10 Assignment/ Semenar-05 Total Marks-15	Better marks out of the two Tot Quiz + obtained marks in Assignment shall be considered against 15 Marks

End Semester Exam (ESE):35	Laboratory/Field Skill Performance: On spot Assessment Section A : Performed the Task based on lab, work 20*1=20 Marks B: Performed the Task based on lab, work (written) 10*1=10Marks Section B : Viva-voce (based on principle/technology) - 5*1=05 Marks

Signature of Convener & Members (CBoS)

SHRI DAVARA UNIVERSITY NAYA RAIPUR

FOUR YEAR UNDERGRADUATE PROGRAM (2024-28)

DEPARTMENT OF PHYSICS

COURSE CURRICULUM

PART-A: Introduction			
Program: Bachelor in Sciences (Certificate/Diploma/Degree Honors)		**Semester-1**	**Session: 2024-2025**
Course Code	**PHSC-01T**		
Course Title	**Mechanics**		
Course Type	**Discipline Specific course (DSC)**		
Pre-requisite(if any)	**As per program**		
Course Learning. Outcomes (CLO)	**At the end of this course, the students will be able to-** ➢ **Analyze and apply the laws of motion to various dynamical situations.** ➢ **Explain and demonstrate the principle of conservation of momentum and energy including their application in real-world scenario such as collision and energy transformation.** ➢ **Evaluate and calculate moment of inertia for objects of different shapes and analyze how these properties affect the motion of rotating bodies.** ➢ **Analyze flow of fluids. Describe special relativistic effects and their effects on the mass and energy of a moving object.**		
Credit Value	**3 Credits**	**Credit =15 Hours-learning & Observation**	
Total Marks	**Max. Marks:=100**	**Min Passing Marks: 40**	

PART -B: Content of the Course		
Total No. of Teaching-learning Periods (01 Hr. per period) -45 Periods (45 Hours)		
Unit	Topics (Course contents)	
I	**Historical Background**: Contribution of Aryabhatta and Varahmihir to science and society, Brief biography of Vikram Sarabhai with his contribution. **Vectors:** Scalar and vector quantities &fields, Scalar & Vector products of two vectors, Derivatives of a vector, Gradient of scalar field and its physical significance**.** **Laws of Motion**: Review of Newton's Laws of motion, Dynamics of a system of particles, Concept of Center of Mass, Motion of center of mass, Conservation of linear momentum, **Work and Energy:** Work-Energy theorem for conservative forces, Force as a gradient of Potential Energy, Conservation of energy, Elastic and in-elastic Collisions	12
II	**Rotational Dynamics:** Angular momentum, Torque, Conservation of angular momentum, Moment of Inertia, Theorem of parallel and perpendicular axes(statements only), Calculation of Moment of Inertia of discrete and continuous objects (Rectangular lamina, dise, solid cylinder, solid sphere).	11

	Elasticity: Stress & Strain, Hooke's law, Elastic constants, Poisson's Ratio, Work done in twisting a cylinder. Fluid **Dynamics:** Flow of fluids, Coefficient of viscosity, Derivation of Poiseulli's formula, Motion of spherical body falling in a viscous fluid, Stoke's law,	
III	**Gravitation:** Newton's Law of Gravitation, Motion of a particle in a central force field (motion is in a plane, angular momentum is conserved, areal velocity is constant), Kepler's Laws (statements. only), Satellite in circular orbit and applications, Geosynchronous orbits. **Oscillations:** Simple harmonic motion, Differential equation of SHM and its solutions, Kinetic and Potential Energy, Total Energy and their time averages, Compound pendulum,	11
IV	**Special Theory of Relativity:** Frame of reference, Galilean Transformations, Inertial and Non- inertial frames, Outcomes of Michelson Morley's Experiment, Postulates of Special Theory of relativity, Lorentz Transformation, Length contraction, Time dilation, Relativistic transformation of velocity, Relativistic variation of mass, Mass-energy equivalence, Transformation of Energy and Momentum	11
Keywords	Aryabhatta, Vectors, Newton's Laws, Angular Momentum, Elasticity, Gravitation, Oscillations, Relativity	
Signature of Convener & Members (CBoS)		

SHRI DAVARA UNIVERSITY NAYA RAIPUR

FOUR YEAR UNDERGRADUATE PROGRAM (2024-28)

DEPARTMENT OF PHYSICS

COURSE CURRICULUM

FOUR YEAR UNDERGRADUATE PROGRAM (2024-28)

PART-C: Learning Resources

Text Books, Reference Books and Others

. Advanced Practical Physics for Students by B.L. Worsnop and H.T. Flint
2. Practical Physics by G.L. Squires
3. An Introduction to Error Analysis: The Study of Uncertainties in Physical Measurements by John R. Taylor
4. Mechanics and Properties of Matter by J.C. Upadhyaya

Text Books Recommended-

- Advanced Practical Physics for students, B.L.Flint&H.T. Worsnop, 1971, Asia Publishing House.
- Engineering Practical Physics, S.Panigrahi& B.Mallick, 2015, Cengage Learning India Pvt. Ltd.
- A Text Book of Practical Physics, Indu Prakash and Ramakrishna, 11th Edition, 2011, Kitab Mahal, New Delhi.
- Practical Physics B.Sc. 1: R P Goyal, Shivlal Publications

Online Resources-

- e-books and e-learning portals
- http://www.swayam.ac.in
- http://www.ignou.ac.in
- http://www.egvankosh.ac.in
- http://www.itm.sc.in
- http://www.eskillindia.org
- http://www.eshiksha.mp.gov.in
- http://www.viah.co.in
- http://www.internshala.com

Online Resources-
e-sources/e-books and e-learning portals

- https://www.pbs.org/video/botany-basics-iuu2bl/
- https://efaidohmannibpcapcalclefindorkaj/https://www2.ca.uky.edu/apcom/pubs/ho/ho96/ho96.pdf
- https://www.botanytoday.com/brunches-of-botany

PART -D: Assessment and Evaluation

Suggested Continuous Evaluation Methods:
Maximum Marks: 100 Marks
Continuous Internal Assessment (CIA): 30 Marks
End Semester Exam (ESE): 70 Marks

Continuous Internal Assessment (CIA): 30 (By Course Teacher)	Internal Test/Quiz:20+20 Assignment/ Semenar-10 Total Marks-30	Better marks out of the two Tot Quiz + obtained marks in Assignment shall be considered against 15 Marks

End Semester Exam (ESE):70	Two section A&B Section A :Q1 Objective 10*1=10 Marks, Q2 Short answer type-5*4=20 Section B : Descriptive answer type qts 1 out of 2frm each- 4*10=40 Marks

Signature of Convener & Members (CBoS)

SHRI DAVARA UNIVERSITY NAYA RAIPUR

DEPARTMENT OF PHYSICS

COURSE CURRICULUM

PART- A: Introduction			
Program: Bachelor in Sciences **(Certificate/Diploma/Degree/Honors**		**Semester-1**	**Session: 2024-2025**
Course Code	**PHSC-01P**		
Course Title	**Lab. Course -01 Mechanics**		
Course Type	**Laboratory course**		
Pre-requisite(if any)	**As per program**		
Course Learning. Outcomes (CLO)	**At the end of this course, the students will be able to-** ➢ **Assemble required parts/devices and arrange them to perform experiments.** ➢ **Record/ observe data as required by the experimental objectives.** ➢ **Analyze recorded data and formulate it to get desired results. Interpret results and check for attainment of proposed objectives related to laws of mechanics and its applications.**		
Credits Value	**1 Credits**	**Credit =30 Hours Laboratory or Field learning/Training**	
Total Marks	**Max. Marks:50**	**Min Passing Marks: 20**	
PART-B:	**Content of the Course**		
Total No. of learning-Training/performance Periods: 30 Periods (30 Hours)			
Module	Topics (Course contents)		No. of Period
Lab/ field Training/ Experiment Content of Course	1. . Measurements of length (or diameter) using vernier caliper, screw gauge and travelling microscope. 2. To study the random error in observations. 3. To study the motion of the spring and calculate (a) Spring constant and, (b) g. 4. To determine the Moment of Inertia of a Flywheel. 5. To determine g and velocity for a freely falling body using Digital Timing Technique. 6. To determine Coefficient of Viscosity of water by Capillary Flow Method (Poiseuille's method). 7. To determine the Young's Modulus of a Wire by Optical Lever Method 8. To determine the Modulus of Rigidity of a Wire by Maxwell's needle. 9. To determine the elastic constants of a wire by Searle's method. 10. To determine the value of g using Bar Pendulum. 11. To determine the value of g using Kater's Pendulum. 12. Study of bending of a beam/ cantilever. To determine Moment of Inertia of an irregular body by Inertia Table.		30
Keywords	Moment of Inertia , Pendulum, Vernier Callipers, Screw Gauge, Travelling microscope, Elastic Constant		
Signature of Convener & Members (CBoS)			

SHRI DAVARA UNIVERSITY NAYA RAIPUR

FOUR YEAR UNDERGRADUATE PROGRAM (2024-28)

DEPARTMENT OF PHYSICS

COURSE CURRICULUM

PART-C: Learning Resources
Text Books, Reference Books and Others
Text Books Recommended-
1. Advanced Practical Physics for students, B.L.Flint&H.T. Worsnop, 1971, Asia Publishing House. 2. Engineering Practical Physics, S.Panigrahi& B.Mallick, 2015, Cengage Learning India Pvt. Ltd. 3. A Text Book of Practical Physics, Indu Prakash and Ramakrishna, 11th Edition, 2011, Kitab Mahal, New Delhi. 4. Practical Physics B.Sc. 1: R P Goyal, Shivlal Publications
Reference Books Recommended- 1. Advanced Practical Physics for Students by B.L. Worsnop and H.T. Flint 2. Practical Physics by G.L. Squires 3. An Introduction to Error Analysis: The Study of Uncertainties in Physical Measurements by John R. Taylor 4. Mechanics and Properties of Matter by J.C. Upadhyaya
Online Resources- ➢ E-resources/e-books and e-learning portals ➢ http://www.swayam.ac.in ➢ http://www.ignou.ac.in ➢ www.egyankosh.ac.in ➢ www.litm.ac.in ➢ www.eskillindia.org ➢ www.eshiksha.mp.gov.in ➢ www.vlab.co.in
Online Resources- e-sources/e-books and e-learning portals ➢ https://www.ncbi.nlm.nih.gov/pmc/articles/PMC5871155/ ➢ https://cms.botany.org/home/careers-jobs/careers-in-botany/arcas-of-specialization- in-botany.html
PART -D: Assessment and Evaluation
Suggested Continuous Evaluation Methods: Maximum Marks: 50 Marks Continuous Internal Assessment (CIA): 15 Marks End Semester Exam (ESE): 35 Marks

Continuous Internal Assessment (CIA): 15 (By Course Teacher)	Internal Test/Quiz:10+10 Assignment/ Semenar-05 Total Marks-15	Better marks out of the two Tot Quiz + obtained marks in Assignment shall be considered against 15 Marks

SHRI DAVARA UNIVERSITY NAYA RAIPUR

End Semester Exam (ESE):35	Laboratory/Field Skill Performance: On spot Assessment Section A : Performed the Task based on lab, work 20∗1=20 Marks B: Sporting based on lab, work (written) 10*1=10Marks Section B : Viva-voce (based on principle/technology) - 5*1=05 Marks	
Signature of Convener & Members (CBoS)		

SHRI DAVARA UNIVERSITY NAYA RAIPUR

FOUR YEAR UNDERGRADUATE PROGRAM (2024-28)

DEPARTMENT OF ZOOLOGY

COURSE CURRICULUM

PART-A: Introduction			
Program: Bachelor in Sciences (Certificate/Diploma/Degree Honors)		**Semester-1**	**Session: 2024-2025**
Course Code	**ZOSC-01T**		
Course Title	**Life on Earth and Unique Attributes of Animal Kingdom**		
Course Type	**Discipline Specific course (DSC)**		
Pre-requisite(if any)	**As per program**		
Course Learning. Outcomes (CLO)	➢ **After successfully completing this course, the students will be able to -** ➢ **Development an understanding of concept mechanism evolutionary signification and relevance of origin of life.** ➢ **Understand General Idea about Invertebrate and Vertebrate animals with special reference and their specific qualities.** ➢ **Understand and appreciate diversity of life form.** ➢ **Apply the knowledge about animals Sciences in daily life.**		
Credit Value	**3 Credits**	**Credit =15 Hours-learning & Observation**	
Total Marks	**Max. Marks:=100**	**Min Passing Marks: 40**	

PART -B: Content of the Course		
Total No. of Teaching-learning Periods (01 Hr. per period) -45 Periods (45 Hours)		
Unit	Topics (Course contents)	
I	**Origin of life: Theories of Origin of life:** Ancient Theory Theory of Special Creation (Mythological approach), Theory of Panspermia or Cosmogonic Theory, Theory of Directed Panspermia, Theory of Catastrophism, Theory of Spontaneous Generation (Abiogenesis or Autogenesis), Theory of Biogenesis: Redi's Experiment and Pasture's Experiment. Modern Theory: Origin of Universe: Big Bang Hypothesis in Brief, Origin of Solar System and The Earth: Nebular hypothesis, Atmosphere and Energy 12 Sources on Primitive Earth, Biochemical Origin of Life: Oparin and Haldane Theory, Chemogenic: Formation of simple and compels organic compounds (Stanely Miller and Ure's Experiment), Formation of Coacervates, Nucleic Acids. Biogeny: Origin of primitive prokaryotic cell. Evolution of modes of Nutrition: Chemoheterotrophs, Anaerobic and Aerobic Photoautotrophic. Evolution of Eukaryotes.	12
II	**Systematics & Unique attributes of Invertebrate and Vertebrate animals with special reference to Coelentrata, Mollusca and Pisces:** Definition and difference between Invertebrate and Vertebrate. Nomenclature: Binomial and Trinomial Nomenclature and International code of Nomenclature Corals: Meaning of Coral, Structure of Coral polyp, Coral Skeleton, Types of corals: Hydrozoan Coral, Example Millipora, Octocorallian Coral, Example Alcyonium, Hexacorallian Corals, Example Gorgonia. **Torsion in Mollusca:** Definition, Mechanism of Torsion, Effects of Tersion, Significance of Tonion. **Pisces: Migration in fishes:** Catadromous: Eel fish and Anadromour Salmon fish **and Parental care in fishes:** By nest formation, Coiling round eggs, Attachunent to body ,Integumentary cups, Shelter in mounth,Brood pouch, Mermaids purses,Vivaparity.	11
III	**Unique attributes of Vertebrate animals with special reference to Amphibia & Reptilia: Parental care in Amphibia:** by Nest, by Nursery or Shelter and by Parents	11

	Neoteny in Amphibia: Definition, Partial and Total Neotony, Factors Affecting Neotony, Examples Axolotal larva, Necturus and Siren. **Reptilia: Venomous & Non-venomous Snakes:** Identification, Poison apparatus: Poison Glands, Poison ducts and Fangs, Biting Mechanism.Snakes: Identification, Poison apparatus: Poison Glands, Poison ducts and Fangs, Biting Mechanism.	
IV	**Unique attributes of Vertebrate animals with special reference to Aves and Mammals: Birds:** Fligle Adaptation, Migration and Perching Mechanism, Flightless Birds (Morphology and Special Characters of Emu, Ostrich and Penguina), Discuss-Birds are glorifled reptiles Archaeopterys. **Monotremes or Egg laying mammals:** Morphology and Special Characters of Echidna and Duck bill platypus. **Aquatic Mammals:** Morphology and Special Characters of Whale and Dolphin **Mammals: Flying Mammals:** Morphology and Special Characters of Bat.	11
Keywords	Origin of life, Invertebrate Vertebrate Corals, Torsion parental care, Neotony Fangs, Aves, Mammals.	
Signature of Convener & Members (CBoS)		

SHRI DAVARA UNIVERSITY NAYA RAIPUR

FOUR YEAR UNDERGRADUATE PROGRAM (2024-28)

DEPARTMENT OF ZOOLOGY

COURSE CURRICULUM

PART-C: Learning Resources

Text Books, Reference Books and Others

Text Books Recommended-

1. E. J. W. Barrington, Invertebrate structure and function, English Language Book Society UK.
2. Robert Barnes, Invertebrate Zoology, Robert Barnes IVth edition Holt Saunders International Edition Japan.
3. Park Haswell, Marshall and Williams, A textbook on Zoology Invertebrate, AITBS Publishing and Distributers, Delhi.
4. Park Haswell, Marshall and Williams, A textbook on Zoology Vertebrate, ATTBS Publishing and Distributers, Delhi.

Reference Books Recommended-

1. ProfR. L. Kotpal, Protozoa to Echinodermata, Rastogi Publication Meerut.
2. EL. Jordan, Dr. P. S. Verma, Invertebrate Zoology, S. Chand Publications, New Delhi.
3. N. Arumugam, N. C. Nair S. - Invertebrate Zoology, Saras Publication..
4. Barrington E. J. W., Invertebrate Structure and Function, Nelson London.
5. Barnes, R. D., Invertebrate Zoology-Saunders Philadelphia.
6. R. L. Kotpal, Invertebrate, Rastogi Publications R. I.. Kotpal, Vertebrate, Rastogi Publications.
7. H. S. Bhampah, KavitaJuneja, Recent trends in vertebrates vol 1-9, Anmol Publication.
8. S. N. Prasad, Life of invertebrates, Vikash Publication House Pvt Ltd New Delhi.
9. G. S. Sandhu, Harshwardhan Bhagskar-Advanced invertebrate zoology-Campus books international.

Online Resources-

- e-books and e-learning portals
- https://www.coursera.org/lecture/emergence-of-life/4-5-invertebrates-successes-of-life-http://www.ignou.ac.in
- https://www.shiksha.com/online-courses/introduction-to-biology-biodiversity-course-http://www.itm.sc.in
- https://www.youtube.com/watch?v=uK-XYhttp://www.eshiksha.mp.gov.in
- https://www.youtube.com/watch?v=WxMSckEcio4http://www.internshala.com

Online Resources-
e-sources/e-books and e-learning portals

- https://www.pbs.org/video/botany-basics-iuu2bl/
- https://efaidohmannibpcapcalclefindorkaj/https://www2.ca.uky.edu/apcom/pubs/ho/ho96/ho96.pdf
- https://www.botanytoday.com/brunches-of-botany

PART -D: Assessment and Evaluation

Suggested Continuous Evaluation Methods:
Maximum Marks: 100 Marks
Continuous Internal Assessment (CIA): 30 Marks
End Semester Exam (ESE): 70 Marks

Continuous Internal Assessment (CIA): 30 (By Course Teacher)	Internal Test/Quiz:20+20 Assignment/ Semenar-10 Total Marks-30	Better marks out of the two Tot Quiz + obtained marks in Assignment shall be considered against 15 Marks

End Semester Exam (ESE):70	Two section A&B Section A :Q1 Objective 10*1=10 Marks Q2 Short answer type-5*4=20 Section B : Descriptive answer type qts 1 out of 2frm each- 4*10=40 Marks

Signature of Convener & Members (CBoS)

SHRI DAVARA UNIVERSITY NAYA RAIPUR

FOUR YEAR UNDERGRADUATE PROGRAM (2024-28)

DEPARTMENT OF ZOOLOGY

COURSE CURRICULUM

PART- A: Introduction			
Program: Bachelor in Sciences **(Certificate/Diploma/Degree/Honors**		**Semester-1**	**Session: 2024-2025**
Course Code	**BOSC-01P**		
Course Title	**Lab. Course -01 (Life on Earth and Unique Attributes of Animal Kingdom)**		
Course Type	**Laboratory course**		
Pre-requisite(if any)	**As per program**		
Course Learning. Outcomes (CLO)	➢ **After successfully completing this course, the students will be able to -** ➢ **To demonstrate comprehensive understanding of the current theories and hypotheses regarding the origin of life on Earth.** ➢ **Understand some distinctive invertebrate and vertebrate animals.** ➢ **Apply this understanding to Broader context of life.**		
Credits Value	**1 Credits**	**Credit =30 Hours Laboratory or Field learning/Training**	
Total Marks	**Max. Marks:50**	**Min Passing Marks: 20**	

PART-B:	**Content of the Course**	
Total No. of learning-Training/performance Periods: 30 Periods (30 Hours)		
Module	Topics (Course contents)	No. of Period
Lab/ field Training/ Experiment Content of Course	1. Study of origin of life through chart and models. 2. Study of different Invertebrates and Vertebestes animals through models and museum specimens in the Jabonstory with details of bie biegeography and diagnostic features: Millipora, Alcyonium, Gorgonia, Hippocampus, Ichthyophis (Female), Alytes (Male), Axolotal larva, Necturus, Siren, Cobra, Viper (pit & Pitlens), Sea Snake, Rattle Snake, Archaeopteryx, Enn, Ostrich and Penguins, Echidna and Duck bill platypus, Whale, Dolphin, Bat. 3. Preparation and Demonstration of Key for Identification of Venomous and Non-venomous snakes. 4. Study of Coral Reify through Models, Photographs. 5. Study of Fossils through chart/ Models. 6. An "Animal album or Practical Record" containing sketches, photograph cut outs, with appropriate write up about the above mentioned taxa. 7. Study of some videos to develop understanding and acquired knowledge on the animals salient features catures as mentioned above. 8. Group discussion/Viva or Seminar presentation on related topic mentioned in Theory paper.	30
Keywords	*Museum specimens, Invertebrates, Vertebrates, Venomous and Non venomous, Seminar.*	
Signature of Convener & Members (CBoS)		

SHRI DAVARA UNIVERSITY NAYA RAIPUR

FOUR YEAR UNDERGRADUATE PROGRAM (2024-28)

DEPARTMENT OF ZOOLOGY

COURSE CURRICULUM

PART-C: Learning Resources

Text Books, Reference Books and Others

Text Books Recommended-

1. S.S. Lal, Practical Zoology, Invertebrate. 12 Edition Rastogi Publications, Meerut, New Delhi.
2. A manual of practical Zoology. Dr. P.S Verma, S. Reference Books Recommended- Chand Publication, New Delhi.

Reference Books Recommended-

1. Park Haswell, Marshall and Williams, A textbook on Zoology Invertebrate, AITBS Publishing and Distributers, Delhi .
2. Park Haswell, Marshall and Williams, A textbook on Zoology Vertebrate, AFTBS Publishing and Distributers, Delhi.

Online Resources-

- E-resources/e-books and e-learning portals
- http://ndi.ätkgp.ac.in/he/document/swayamprabha/swayam
- http://www.swayam.ac.in
- http://www.ignou.ac.in
- www.egyankosh.ac.in
- www.litm.ac.in
- www.eskillindia.org
- www.eshiksha.mp.gov.in

Online Resources-
e-sources/e-books and e-learning portals

- https://www.ncbi.nlm.nih.gov/pmc/articles/PMC5871155/
- https://cms.botany.org/home/careers-jobs/careers-in-botany/arcas-of-specialization- in-botany.html

PART -D: Assessment and Evaluation

Suggested Continuous Evaluation Methods:
Maximum Marks: 50 Marks
Continuous Internal Assessment (CIA): 15 Marks
End Semester Exam (ESE): 35 Marks

Continuous Internal Assessment (CIA): 15 (By Course Teacher)	Internal Test/Quiz:10+10 Assignment/ Semenar-05 Total Marks-15	Better marks out of the two Tot Quiz + obtained marks in Assignment shall be considered against 15 Marks

End Semester Exam (ESE):35	Laboratory/Field Skill Performance: On spot Assessment Section A : Performed the Task based on lab, work 20*1=20 Marks B: Spotting frased on tools & technology (written) 10*1=10Marks Section B : Viva-voce (based on principle/technology) - 5*1=05 Marks

Signature of Convener & Members (CBoS)

SHRI DAVARA UNIVERSITY NAYA RAIPUR

FOUR YEAR UNDERGRADUATE PROGRAM (2024-28)

DEPARTMENT OF POLITICAL SCIENCES

COURSE CURRICULUM

PART-A: Introduction			
Program: Bachelor in Sciences (Certificate/Diploma/Degree Honors)		**Semester-1**	**Session: 2024-2025**
Course Code	**PSGE-01**		
Course Title	**Introduction to Political Theory**		
Course Type	**Discipline Specific course (GE)**		
Pre-requisite(if any)	**As per program**		
Course Learning. Outcomes (CLO)	➢ **Create the understanding of the concept of political science, and methodology.** ➢ **Evaluate the concept of state, Its theories of norm, functions and relation with individuals.** ➢ **Analyses the basic concepts of Political Science like liberty, right, sovereignty.** ➢ **Apply the knowledge of democracy and democratic norms, the functional machinery of electoral democracy like political party system and pressure groups Rule of State as welfare agency, and as an agency of social change.**		
Credit Value	**4 Credits**	**Credit =60 Hours-learning & Observation**	
Total Marks	**Max. Marks:=100**	**Min Passing Marks: 40**	

PART -B: Content of the Course		
Total No. of Teaching-learning Periods (01 Hr. per period) -45 Periods (45 Hours)		
Unit	Topics (Course contents)	
I	**Political Science -Initial** Political science Concept, nature, Scope Power, Authority meaning, characteristics, types. Legitimacy concept, relationship of power, authority and legitimacy Study methods of political science, Behaviouraliam and post-behaviouralism.	12
II	**State** State: Concept, Development of Stine, Essential Elements Theories of origin state-Divine, power theory, social contract and evolutionary theory, Theories of functions of state-Marxast, liberal, neo-liberal, pluralist, theory Law Definition Source, Classification Public welfare state Nationalism. Concept, types.	11
III	**Concept** Sovereignty concept, types, Characteristics. Principles Meaning, types major Theories, Duties. Freedom Meaning Types, Positive and Negative Theory of Freedom Equality Meaning type and relation to freedom Political Obligation, Justice Concept, types. Democracy Concept, types, Mints and dements, Principles of democracy Necessary conditions by the success of Democracy.	11
IV	**State in Function** Forms of Geversument Unitary and Federal, Parliamentary and Pressdemial Totalitarianum Concept, types	11

	Organs of Governinent Legislature, Executive and Judiciary Theory of Separation of Powers and Checks and Balances Constitution meaning and Lands Petincal Party meaning, knals, majus thences, merits and demerits Pressure Groups meaming, kunds and techinoque Public Opinion Social hatice, Theones of Representation	
Keywords	Political theory, sate, sovereignty, right, they, democracy, constitution, party.	
Signature of Convener & Members (CBoS)		

SHRI DAVARA UNIVERSITY NAYA RAIPUR

FOUR YEAR UNDERGRADUATE PROGRAM (2024-28)

DEPARTMENT OF POLITICAL SCIENCE

COURSE CURRICULUM

PART-C: Learning Resources

Text Books, Reference Books and Others

Text Books Recommended-

1. Ambadatt Pant Harimohan Jain Madan Gopal (1985) Fundamentals of Political Science, Central Publishing House Allahabad. U.P.
2. Sandhu Man Singh (1956) Political Theory Hindi Medium Implementation Directorate, Delhi University, New Delhi
3. Johari JC 1916) Basic principles of political science, Sahitya Bhavan, Agra.
4. Rajeev and Ashok Acharya (Eds) Political Theory A Flag, Dilsey Pearson, 2008

Reference Books Recommended-

1 umar, Sanjeev (Ed. Understanding of Political Theory, Delhi: Orient Book Swan, 2019
2 Hussain Shakeel (2018) Conceptual Introduction to Political Theory. Chhattisgarh State Hindi Forest Academy, Rampur.
3 K.K. Mishra (2010) Political Theory, 5. Chand Publishing Delhi
4 OP Gouba (2014) An Introduction to Political Theory, MacMillan Publishers, Delhi

Online Resources-

- e-books and e-learning portals
- https://www.coursera.org/lecture/emergence-of-life/-http://www.ignou.ac.in
- https://www.shiksha.com/online-courses/-http://www.itm.sc.in
- https://www.youtube.com/watch?v=uK-XYhttp://www.eshiksha.mp.gov.in
- https://www.youtube.com/watch?v=WxMSckEcio4http://www.internshala.com

Online Resources-
e-sources/e-books and e-learning portals

- https://www.pbs.org/video/political-basics-iuu2bl/
- https://efaidohmannibpcapcalclefindorkaj/https://www2.ca.uky.edu/apcom/pubs/ho/ho96/ho96.pdf
- https://www.botanytoday.com/brunches-of-botany

RT -D: Assessment and Evaluation

Suggested Continuous Evaluation Methods:
Maximum Marks: 100 Marks
Continuous Internal Assessment (CIA): 30 Marks
End Semester Exam (ESE): 70 Marks

Continuous Internal Assessment (CIA): 30 (By Course Teacher)	Internal Test/Quiz:20+20 Assignment/ Semenar-10 Total Marks-30	Better marks out of the two Tot Quiz + obtained marks in Assignment shall be considered against 15 Marks

End Semester Exam (ESE):70	Two section A&B Section A :Q1 Objective 10∗1=10 Marks Q2 Short answer type-5∗4=20 Section B : Descriptive answer type qts 1 out of 2frm each- 4*10=40 Marks

Signature of Convener & Members (CBoS)

SHRI DAVARA UNIVERSITY NAYA RAIPUR

FOUR YEAR UNDERGRADUATE PROGRAM (2024-28)

DEPARTMENT OF ENGLISH

COURSE CURRICULUM

PART-A: Introduction			
Program: Bachelor in Sciences (Certificate/Diploma/Degree Honors)		**Semester-1**	**Session: 2024-2025**
Course Code	**AEC-01**		
Course Title	**Communicative English and Soft Skills**		
Course Type	**Ability Enhancement Course**		
Pre-requisite(if any)	**As per program**		
Course Learning. Outcomes (CLO)	**After the completion of this course, the students will be able to-** ➢ **Understand and apply the use of Articles and Tenses in day to daylife** ➢ **Analyze the power of imagination and creativity and critically appreciate the poems.** ➢ **Identify and develop different types of writing skills.** ➢ **Appreciate and value the use of idioms and phrases as enriching elements of language expression.**		
Credit Value	**2 Credits**	**Credit =30 Periods -learning & Observation**	
Total Marks	**Max. Marks:=50**	**Min Passing Marks: 20**	

PART -B: Content of the Course		
Total No. of Teaching-learning Periods (45 Min. per period) -30 Periods		
Unit	Topics (Course contents)	
I	**Prose:-** **1**. Darshana Dholakia: Baa-My Mother-A Person, A Woman. 2. Anita Desai: A Devoted Son. 3. Rabindranath Tagore: The Home Coming.	08
II	**Poetry:** 1. William Wordsworth: The Solitary Reaper 2. Robert Lee Frost: Stopping by the Woods on a Snowy Evening	07
III	**Letter Writing:** 1. Formal Letter 2. Informal Letter **Composition:** 1. Describing a Place or a Person 2. Writing a Biographical Sketch 3. Narrating an Event or Experience.	08
IV	**Writing Skills:** 1. Word Formation, Idioms and Phrases 2. Coordination and Subordination, One Word Substitutes **Grammar**: 1. Articles 2. Tenses	07
Keywords	Political theory, sate, sovereignty, right, they, democracy, constitution, party.	
Signature of Convener & Members (CBoS)		

SHRI DAVARA UNIVERSITY NAYA RAIPUR

FOUR YEAR UNDERGRADUATE PROGRAM (2024-28)

DEPARTMENT OF ENGLISH

COURSE CURRICULUM

PART-C: Learning Resources

Text Books, Reference Books and Others

Text Books Recommended-

1. Fluency in English - Part 11, Oxford University Press, 2006.
2. Enrich Your English, OUP, SR Inthira and V. Saraswathi, CIEFL, 1997
3. Oxford A-Z of English Usage, ed. Jeremy Butterfield, OUP, 2007.

Reference Books Recommended-

1. Longman Dictionary of Common Errors, N.D. Turton and J.B. Heaton, Longman, 1998.
2. Contemporary Communicative English, S Chand
3. Malhotra Prerna, Deb Dulal Halder, (2019) Communication Skills: Theory and Practice, Eighth Edition, Book Age Publications, New Delhi.

Online Resources-

- Applying Communication Theory for Professional Life: A Practical Introduction. Dainton and Zelley, http://taime.uz.ac.zw/claroline/backends/download.php?urf-L0ludHJvX3RvX2NvbW1lbmljYXRpb25f
- https://www.coursera.org/lecture/emergence-of-life/4-5-invertebrates-successes-of-life-http://www.ignou.ac.in
- https://web.sol.du.ac.in/my modules/type/cbcs-11- 2/data/root/B.Com/Semester%202/ABILITY-ENHANCEMENT%20COMPULSORY%20COURSE-AECC/English%20Communication%20A-B-C/Unit%201-5.
  - pdf https://www.youtube.com/watch?v=uK-XYhttp://www.eshiksha.mp.gov.in
  - https://www.youtube.com/watch?v=WxMSckEcio4http://www.internshala.com
  - https://archive.org/details/personality-development-book/mode/lup
  - https://www.coursera.org/articles/presentation-skills
  - https://www.cbs.de/en/blog/15-effective-presentation-tips-to-improve-presentation-skills/
  - https://benjaminball.com/blog/good-body-language-best-visual-aid-talks/
  -

Online Resources-

- e-sources/e-books and e-learning portals https://blog.moderngov.com/importance-of-body-language-in-presentations-good-bad-
- https://efaidohmannibpcapcalclefindorkaj/https://www2.ca.uky.edu/apcom/pubs/ho/ho96/ho96.pdf
- https://www.botanytoday.com/brunches-of-botany

PART -D: Assessment and Evaluation

Suggested Continuous Evaluation Methods:
Maximum Marks: 50 Marks
Continuous Internal Assessment (CIA): 15 Marks
End Semester Exam (ESE): 35 Marks

Continuous Internal Assessment (CIA): 15 (By Course Teacher)	Internal Test/Quiz:10+10 Assignment/ Semenar-05 Total Marks-15	Better marks out of the two Tot Quiz + obtained marks in Assignment shall be considered against 15 Marks

End Semester Exam (ESE):70	Two section A&B Section A :Q1 Objective 1*5=5 Marks Q2 Short answer type-2*5=10 (I. Vocabulary, II Unseen passage Section B : Descriptive answer type qts 1 out of 2frm each- 5*4=20 Marks

Signature of Convener & Members (CBoS)

SHRI DAVARA UNIVERSITY NAYA RAIPUR

FOUR YEAR UNDERGRADUATE PROGRAM (2024-28)

DEPARTMENT OF COMPUTER SCIENCES

COURSE CURRICULUM

PART-A: Introduction			
Program: Bachelor in Sciences (Certificate/Diploma/Degree Honors)		**Semester-1**	**Session: 2024-2025**
Course Code	**VAC-01**		
Course Title	**Computer fundamental & MS Office**		
Course Type	**Value Additional Course**		
Pre-requisite(if any)	**As per program**		
Course Learning. Outcomes (CLO)	**Completing this course, students will be able to: -** **3. Study and use of basic concepts and terminology of information technology.** **4. Organize files and documents on storage devices.** **5. Acquire knowledge of ICT and Internet applications.** **6. Develop information technology solutions by evaluating user requirements in advance trends of IT.** **7. Acquire knowledge of MS-Excel, MS-PowerPoint and MS-Access.**		
Credit Value	**1 Credits**	**Credit =30 Periods -learning & Observation**	
Total Marks	**Max. Marks:=50**	**Min Passing Marks: 20**	

PART -B: Content of the Course		
Total No. of Teaching-learning Periods (45 Min. per period) -15 Periods		
Unit	Topics (Course contents)	
I	**Introduction to Computer:** History of computer, Generations and Classification, Basic Anatomy of Computer Block Diagram, Central Processing Unit (CPU): Function of each Unit, Memory: Primary, Cache, Flash, Software and its needs, Types of S/W: System Software and Application Software, Types of Programming Language: Machine Language, Assembly Language, High Level Language their advantages and disadvantages, Language Processors/Translators: Assembler, Interpreter and Compiler, Fundamental of Information Technology: Data and Information, Concept of IT, Application of IT, What is ICT?, Components of ICT, Impact of ICT in Society. Advanced Trends in IT: Cloud Technology, Virtual LAN Technology, M-Commerce, Nanotechnology, Virtual Reality, 3-D Printing, Internet of Things (IoT), Artificial Intelligence (AI), Machine Learning (ML), Cloud Computing, Quantum Computing, G- Suite, Gol digital initiatives in higher education: SWAYAM, Swayam Prabha, National Academic Depository, National Digital Library of India, E-Sodh-Sindhu, Virtual labs, e-Yantra and NPTEL	08
II	**MS-Word:** Introduction to word processing software and its features, Creating new document, Saving documents, Opening and Printing documents. Home Tab: Setting fonts. Paragraph settings, Various styles (Normal, No spacing, Heading l. Heading2, Title, Strong), Find & Replace, Format painter, Copy paste and paste special. Insert Tab: Pages, Tables, Pictures, Clipart, Shapes, Header & Footer, Word Art. Equation and Symbols. Page Layout Tab: Page setup, Page Background, Paragraph (indent and spacing). Mailing Tab: Create Envelops and Labels, Mail Merge. Review Tab: Spelling and Grammar check, New comment, Protect document, View Tab: Document views, Zoom, Window (New window, Split, Switch window).	07

III	**MS-Excel: Introducing Excel, Use of Excel sheet, creating new sheet, Saving. Opening.** and Printing workbook. Home Tab: Font, Alignment, Number, Styles and cells and editing, Conditional Formatting. Insert Tab: Table, Charts (column chart, Pie chart, Bar chart, Line chart) and Texts (headąc & footer, word art, signature line), Page Layout Tab: Page setup options, Scale to fit (width, height, scale). Formulas Tab: Auto sum (sum, average, min, max), Logical (IF, and, or, not, true, false), Math & Trig (sin, cos, tan, ceiling, floor, fact, mod, log), Sort and Filter options, Data validation, Group and ungroup. Review Tab: Protect sheet, Protect workbook, and Share workbook. View Tab: Page breaks, Page layout, Freezing Panes, Split and hide.	08
IV	**Working with PowerPoint and MS-Access**: PowerPoint: Introducing PowerPoint, Use of PowerPoint presentation, Creating new slides saving. Opening and printing. Home Tab: New slide, Layout. Reset, Delete, Setting text direction, Align text, Convert to smart art., Drawing options. Insert Tab: Table, Picture, Clipart, Photo album, Smart art, Shapes and chart, Movie and sound. Hyperlink and action, Text box, Word art, Object. Design Tab: Page setup options, Slide orientation, Applying various themes, Selecting background style and formatting it. Animations Tab: Custom animation for entrance, Exit and emphasis, Applying slide transition, Setting transition speed and sound, Animation on rehearse timing. Slideshow & View Tab: Start slide, Show options, and Setup options. View tab: Presentation views, Colors and Window option. **MS-Access:** Introduction to DBMS, features of DBMS, creating blank database Saving it in accdb format, Defining data type in MS Access, Creating tables, creating reports, query wizard.	07
Keywords	Information Technology (IT), Information and Communication Technology (ICT), G-Suit MS Excel, MS Power Point, MS-Access.	
Signature of Convener & Members (CBoS)		

SHRI DAVARA UNIVERSITY NAYA RAIPUR

FOUR YEAR UNDERGRADUATE PROGRAM (2024-28)

DEPARTMENT OF COMPUTER SCIENCE

COURSE CURRICULUM

PART-C: Learning Resources

Text Books, Reference Books and Others

Text Books Recommended-

1. Computer Fundamentals, P.K. Sinha, BPB Publication, Sixth Edition.
2. Fundamentals of Information Technology, Chetan Shrivastava, Kalyan Publishers.
3. Fundamentals of Computers, V. Rajaraman, PHI Sixth Edition.

Reference Books Recommended-

1. Publisher IIP. Computer Fundamentals Architecture and Organization, B. Ram, New Age International Publishers, Fifth Edition.
2. Fundamentals of Information Technology, Alexis Leon and Mathews Leon, Vikash Publication.
3. Introduction to Information Technology, V. Rajaraman, PHI publication.
4. Fundamental of IT, Leon and Leon, Leon Tec world.
5. Introduction to Information Technology, Aksoy and Denardis, Cengage learning.
6. Computers Today, Suresh K. Basandra, Galgotia Publications.
7. Information Technology The breaking wave, Dennis P.Curtin, Kim Foley, Kunai Sen and Cathleen Morin, TMH.

Online Resources-

- Introduction to Computer Fundamental from W3school: https://www.w3schools.blog/computer-fundamentals-tutorial
- Introduction to MS-Word from W3school: https://www.w3schools.blog/ms-word-tutorial
- Introduction to MS-Excel from W3school:https://www.w3schools.com/excel/excel_introduction.php
- ntroduction to MS-PowerPoint from W3school:https://www.w3schools.blog/powerpoint-tutorial Introduction to MS-Access from W3school:
- https://www.youtube.com/watch?v=WxMSckEcio4http://www.internshala.com

Online Resources-
e-sources/e-books and e-learning portals

- https://www.rgycsm.org/uploads/books/MICROSOFT-OFFICE-BOOK.pdf
- https://www.youtube.com/watch?v=SH40YV5AJ6A
- https://www.youtube.com/watch?v=SH40YV5AJ6A
- https://hte.rajasthan.gov.in/dept/dte/board

PART -D: Assessment and Evaluation

Suggested Continuous Evaluation Methods:
Maximum Marks: 50 Marks
Continuous Internal Assessment (CIA): 15 Marks
End Semester Exam (ESE): 35 Marks

Continuous Internal Assessment (CIA): 15 (By Course Teacher)	Internal Test/Quiz:10+10 Assignment/ Semenar-10 Total Marks-15	Better marks out of the two Tot Quiz + obtained marks in Assignment shall be considered against 15 Marks

End Semester Exam (ESE):50	Two section A&B Section A :Q1 Objective 5*1=5 Marks Q2 Short answer type-5*4=20 Section B : Descriptive answer type qts 1 out of 2frm each- 10*1=10 Marks

Signature of Convener & Members (CBoS)

SHRI DAVARA UNIVERSITY

NAYA RAIPUR (C.G.)

PROGRAMME- CURRICULUM

FOR

BECHELOR OF SCIENCES

(Chemistry, Physics, Zoology (PCB)

SEMESTER-II

AS PER NEW EDUCATION POLICY-2020

AND

NATIONAL EDUCATION POLICY-2025

FOUR YEAR UNDERGRADUATE PROGRAMME-2024-25

(EFFECTIVE FROM THE SESSION-2024-2025

SHRI DAVARA UNIVERSITY NAYA RAIPUR

INTRODUCTION OF THE DEPARTMENT: -

Department of Chemistry

Introduction

The Department of Chemistry is a vibrant community of scholars, researchers, and students dedicated to advancing our understanding of the chemical sciences. Our department offers undergraduate and postgraduate programs in chemistry, providing students with a comprehensive education in the principles and applications of chemistry.

Mission

Our mission is to provide students with a rigorous and well-rounded education in chemistry, preparing them for careers in research, industry, education, and beyond. We strive to create an inclusive and supportive environment that encourages collaboration, creativity, and intellectual curiosity.

Research Areas

Faculty and students in the Department of Chemistry engage in cutting-edge research in various areas, including:

1. Organic Chemistry: Synthesis and characterization of organic compounds.
2. Inorganic Chemistry: Study of inorganic compounds and their applications.
3. Physical Chemistry: Investigation of the physical

principles underlying chemical phenomena.

4. Analytical Chemistry: Development and application of analytical techniques.

Department of Physics

Introduction

The Department of Physics is a dynamic community of Mechanics physiologists, researchers, and students dedicated to exploring the fascinating world of physics. Our department offers undergraduate and postgraduate programs in physics, providing students with a comprehensive education in physics.

Mission

Our mission is to provide students with a rigorous and well-rounded education in physics, preparing them for careers in research, conservation, education, and beyond. We strive to create an inclusive and supportive environment that encourages collaboration, creativity, and intellectual curiosity.

Research Areas

Faculty and students in the Department of Physics engage in cutting-edge research in various areas, including:

1. Physics Systematics: Study of physics classification, evolution, and diversity.

2. Physics Mechanism : Investigation of growth,

development, and responses to physics.

Department of Zoology

Introduction

The Department of Zoology is a vibrant community of animal biologists, researchers, and students dedicated to exploring the fascinating world of animals. Our department offers undergraduate and postgraduate programs in zoology, providing students with a comprehensive education in animal biology.

Mission

Our mission is to provide students with a rigorous and well-rounded education in zoology, preparing them for careers in research, conservation, education, and beyond. We strive to create an inclusive and supportive environment that encourages collaboration, creativity, and intellectual curiosity.

Research Areas

Faculty and students in the Department of Zoology engage in cutting-edge research in various areas, including:

1. Animal Systematics: Study of animal classification, evolution, and diversity.
2. Animal Physiology: Investigation of animal growth, development, and responses to environmental stimuli.
3. Animal Ecology: Study of animal interactions with their environment and other organisms.
4. Animal Behavior: Study of animal behavior,

including social behavior, communication, and learning.

VISION OF DEPARTMENT: -

Department of Chemistry

Vision

To be a leading department of chemistry, recognized for its academic excellence, innovative research, and commitment to fostering a community of scholars who can address the complex chemical challenges of the 21st century.

Objectives

1. To provide students with a world-class education in chemistry, emphasizing both theoretical foundations and practical applications.
2. To conduct cutting-edge research in chemistry, focusing on areas of national and global importance.
3. To foster collaborations with other departments, institutions, and industries to promote interdisciplinary research and innovation.
4. To develop and maintain state-of-the-art research facilities and instrumentation.

Department of Physics

Vision

To be a premier department of physics, dedicated to

advancing our understanding of physics and addressing the critical challenges facing our mechanism, including deferments physics branches.

Objectives

1. To provide students with a comprehensive education in physics, emphasizing both theoretical foundations and practical applications.

2. To conduct innovative research in physics, focusing on areas of national and global importance.

3. To develop and maintain a diverse collection of physical data.

4. To foster collaborations with other departments, institutions, and industries to promote interdisciplinary research and innovation.

Department of Zoology

Vision

To be a leading department of zoology, recognized for its academic excellence, innovative research, and commitment to fostering a community of scholars who can address the complex challenges facing animal populations and ecosystems.

Objectives

1. To provide students with a world-class education in zoology, emphasizing both theoretical foundations and practical applications.

2. To conduct cutting-edge research in animal biology,

focusing on areas of national and global importance.

3. To develop and maintain state-of-the-art research facilities and instrumentation.

4. To foster collaborations with other departments, institutions, and industries to promote interdisciplinary research and innovation.

SCOPE OF DEPARTMENT: -

Department of Chemistry

Scope

1. Research and Development: Opportunities exist in various fields like medicinal chemistry, materials science, environmental chemistry, and analytical chemistry.

2. Industry and Manufacturing: Chemistry graduates can work in various industries like pharmaceuticals, petrochemicals, and materials manufacturing.

3. Environmental Conservation: Chemists can work in environmental monitoring, conservation, and sustainability.

4. Education and Academia: Chemistry graduates can pursue teaching and research careers in academic institutions.

5. Government and Policy: Chemists can work in government agencies, regulatory bodies, and policy-making institutions.

Department of Physics

Scope

1. Research and Development: Opportunities exist in various fields like physics.

2. Education and Academia: Physics graduates can pursue teaching and research careers in academic institutions.

5. Government and Policy: Physics can work in government agencies, regulatory bodies, and policy-making institutions.

Department of Zoology

Scope

1. Research and Development: Opportunities exist in various fields like animal physiology, ecology, evolution, and conservation biology.

2. Wildlife Conservation and Management: Zoology graduates can work in wildlife conservation, management, and research institutions.

3. Animal Health and Veterinary Science: Zoologists can work in animal health, veterinary science, and animal biotechnology industries.

4. Education and Academia: Zoology graduates can pursue teaching and research careers in academic institutions.

5. Government and Policy: Zoologists can work in government agencies, regulatory bodies, and policy-making institutions.

PROGRAMME OUTCOME: -

1**. Knowledge and Understanding:** Demonstrate

advanced knowledge and understanding of scientific principles, theories, and concepts in their chosen field.

2. **Critical Thinking and Analysis:** Apply critical thinking and analytical skills to evaluate scientific data, theories, and methodologies.

3. **Research and Problem-Solving:** Design, conduct, and analyze research experiments, and apply scientific principles to solve complex problems.

4. **Communication and Collaboration:** Communicate complex scientific ideas effectively to various audiences, and collaborate with others in a research or professional setting.

5. **Scientific Literacy and Critical Evaluation:** Evaluate the scientific literature, identify gaps in knowledge, and propose new research directions.

6. **Professional Development and Leadership:** Demonstrate leadership skills, manage projects, and develop a professional network in their chosen field.

7. **Knowledge and Understanding:** Demonstrate a solid understanding of scientific principles, theories, and concepts in their chosen field.

8. **Critical Thinking and Analysis:** Apply critical thinking and analytical skills to evaluate scientific data and theories.

9. **Scientific Literacy and Communication:** Communicate scientific ideas effectively to various audiences, and demonstrate an understanding of the scientific method.

10. **Problem-Solving and Laboratory Skills:** Apply scientific principles to solve problems, and demonstrate

laboratory skills and safety protocols.

11. **Teamwork and Collaboration:** Collaborate with others in a laboratory or project setting, and demonstrate an understanding of the importance of teamwork in science.

12. **Preparation for Further Study or Employment:** Demonstrate preparation for further study or employment in a scientific field, and exhibit a commitment to lifelong learning.

COURSE OUTCOME: -

Department of Chemistry

Course Outcomes

1**. Knowledge of Chemical Principles:** Understand and apply fundamental chemical principles, theories, and concepts.

2. **Laboratory Skills:** Develop laboratory skills, including experimentation, data analysis, and safety protocols.

3. **Analytical and Problem-Solving Skills:** Apply analytical and problem-solving skills to solve chemical problems and analyze data.

4. **Communication Skills:** Communicate chemical concepts and research findings effectively through written and oral presentations.

5. Critical Thinking and Scientific Literacy: Evaluate scientific literature, identify gaps in knowledge, and propose new research directions.

SHRI DAVARA UNIVERSITY NAYA RAIPUR

Department of Physics

Course Outcomes

1. **Knowledge of Physical science:** Understand and apply fundamental principles of physics, including deferments branches.

2. **Laboratory and Field Skills:** Develop laboratory and field skills, including plant collection, preservation, and experimentation.

4. Ecological and Environmental Awareness: Understand the importance of physics in ecosystems and the impact of human activities on physical science.

5. **Communication and Critical Thinking:** Communicate physical concepts and research findings effectively and critically evaluate scientific literature.

Department of Zoology

Course Outcomes

1. **Knowledge of Animal Biology:** Understand and apply fundamental principles of animal biology, including morphology, anatomy, physiology, and ecology.

2. **Animal Identification and Classification:** Identify and classify animals using morphological, anatomical, and molecular characteristics.

3. **Laboratory and Field Skills:** Develop laboratory and field skills, including animal collection, preservation, and experimentation.

4. **Ecological and Environmental Awareness:** Understand the importance of animals in ecosystems and

the impact of human activities on animal diversity and ecology.

5. **Communication and Critical Thinking:** Communicate zoological concepts and research findings effectively and critically evaluate scientific literature.

UNIT OUTCOME: -

Department of Chemistry

Unit 1: Atomic Structure and Chemical Bonding

1. Explain the structure of atoms and molecules.

2. Describe the types of chemical bonds and their properties.

3. Apply knowledge of atomic structure and chemical bonding to predict chemical behavior.

Unit 2: Thermodynamics and Kinetics

1. Understand the laws of thermodynamics and their applications.

2. Explain the principles of chemical kinetics and reaction mechanisms.

3. Apply thermodynamic and kinetic principles to solve chemical problems.

Unit 3: Organic Chemistry

1. Identify and classify organic compounds.

2. Explain the mechanisms of organic reactions.

3. Apply knowledge of organic chemistry to synthesize and analyze organic compounds.

Unit 4: Analytical Chemistry

1. Understand the principles of analytical chemistry techniques.

2. Apply analytical techniques to analyze and identify chemical substances.

3. Interpret analytical data to solve chemical problems.

Department of Physics

Mechanics

1. Kinematics: Students will be able to describe the motion of objects in terms of position, velocity, and acceleration.

2. Dynamics: Students will be able to apply Newton's laws of motion to solve problems involving force, mass, and acceleration.

3. Energy and Momentum: Students will be able to analyze the conservation of energy and momentum in physical systems.

Electromagnetism

1. Electric Fields: Students will be able to calculate the electric field due to a point charge and a distribution of charges.

2. Magnetic Fields: Students will be able to calculate the magnetic field due to a current-carrying wire and a magnetic dipole.

3. Electromagnetic Waves: Students will be able to

describe the properties of electromagnetic waves and their applications.

Thermodynamics

1. Temperature and Heat: Students will be able to define temperature and heat, and describe the laws of thermodynamics.

2. Thermodynamic Systems: Students will be able to analyze thermodynamic systems, including ideal gases and heat engines.

3. Entropy and Free Energy: Students will be able to calculate entropy and free energy changes in thermodynamic systems.

Quantum Mechanics

1. Wave-Particle Duality: Students will be able to describe the wave-particle duality of matter and energy.

2. Schrödinger Equation: Students will be able to solve the Schrödinger equation for simple systems, such as the particle in a box.

3. Quantum Systems: Students will be able to analyze quantum systems, including the hydrogen atom and quantum harmonic oscillator.

Statistical Mechanics

1. Probability and Statistics: Students will be able to apply probability and statistical concepts to physical systems.

2. Thermodynamic Systems: Students will be able to

analyze thermodynamic systems using statistical mechanics.

3. Phase Transitions: Students will be able to describe phase transitions and critical phenomena using statistical mechanics.

Computational Physics

1. Numerical Methods: Students will be able to apply numerical methods, such as finite difference and Monte Carlo methods, to solve physical problems.

2. Programming Languages: Students will be able to write programs in languages such as Python, C++, and MATLAB to solve physical problems.

3. Data Analysis: Students will be able to analyze and visualize data using computational tools.

Experimental Physics

1. Experimental Design: Students will be able to design and conduct experiments to test physical hypotheses.

2. Data Analysis: Students will be able to analyze and interpret data from experiments.

3. Instrumentation: Students will be able to use and maintain laboratory instrumentation, such as oscilloscopes and spectrometers.

Department of Zoology

Unit 1: Animal Morphology and Anatomy

1. Identify and describe animal morphological and anatomical features.

2. Explain the functions of animal tissues and organs.

3. Apply knowledge of animal morphology and anatomy to understand animal development and evolution.

Unit 2: Animal Physiology

1. Understand the principles of animal physiology, including nervous and circulatory systems.

2. Explain the mechanisms of animal growth and development.

3. Apply knowledge of animal physiology to solve problems in animal health and welfare.

Unit 3: Animal Ecology

1. Understand the principles of animal ecology, including population dynamics and community structure.

2. Explain the interactions between animals and their environment.

3. Apply knowledge of animal ecology to understand and manage ecosystems.

Unit 4: Animal Genetics and Evolution

1. Understand the principles of animal genetics and evolution.

2. Explain the mechanisms of animal genetic variation and evolution.

3. Apply knowledge of animal genetics and evolution to understand animal diversity and adaptation.

SHRI DAVARA UNIVERSITY NAYA RAIPUR

SEMESTER II

S.NO	COURSE CODE	COURSE TITLE	TEACHING HOURS PER WEEK				EXAMINATION SCHEME				
DISCIPLINE SPECIFIC COURSE (DSC)			L	T	P	C	THEORY		PRACTICAL		TOTAL MARKS
							EX	IN	EX	IN	
1.	CHSC-02T	Fundamental Chemistry-II	2	1	0	3	70	30	-	-	100
2.	PHSC-O2T	Electricity and Magnetism	2	1	0	3	70	30	-	-	100
3.	ZOSC-02T	Cell Biology and Histology	2	1	0	3	70	30	-	-	100
GENERAL ELECTIVE (GE)											
4.	SCGE-02	Constitutional Government in India	3	1	0	4	70	30	-	-	100
ABILITY ENHANCEMENT COURSE (AEC)											
5.	AEC-02	Hindi Language	2	0	0	2	35	15	-	-	50
SKILLS ENHANCEMENT COURSE (SEC)											
6.	SEC-02	Chemistry Lab Skills-II	1	1	0	2	35	15	-	-	50
PRACTICALS (LAB)											
7.	CHSC-02P	Fundamental Chemistry-II	0	0	2	1	-	-	35	15	50
8.	PHSC-O2P	Electricity and Magnetism-LAB	0	0	2	1	-	-	35	15	50
9.	ZOSC-02P	Cell Biology and Histology-LAB	0	0	2	1	-	-	35	15	50
Total Contact hours Per Week:30		Total credit:				20	Total mark				650

SHRI DAVARA UNIVERSITY NAYA RAIPUR

FOUR YEAR UNDERGRADUATE PROGRAM (2024-28)

DEPARTMENT OF CHEMISTRY

COURSE CURRICULUM

PART-A: Introduction			
Programme: Bachelor in Sciences (Certificate/Diploma/Degree Honors)		**Semester-II**	**Session: 2024-2025**
Course Code	**CHSC-02T**		
Course Title	**FUNDAMENTAL CHEMISTRY-II**		
Course Type	**Discipline Specific course (DSC)**		
Pre-requisite (if any)	**As per program**		
Course Learning. Outcomes (CLO)	**At the end of this course, the students will be able to** ➢ **To understand different acid-base theories and solvent system.** ➢ **To learn the preparation, bonding, and reactions of C-C σ- & π-bonded compounds.** ➢ **To understand the concept and chemistry of aromatic compounds and their reactions.** ➢ **To learn the basic concepts of various states of matter & understand the basic concepts of surface chemistry and chemical kinetics.**		
Credit Value	**3 Credits**	**Credit =15 Hours-learning & Observation**	
Total Marks	**Max. Marks:=100**	**Min Passing Marks: 40**	

PART -B: Content of the Course		
Total No. of Teaching-learning Periods (01 Hr. per period) -45 Periods (45 Hours)		
Unit	Topics (Course contents)	
I	**Acid, Base and Solvent System** **Theories of acids and bases: Arrhenius, Bronsted-Lowry, conjugate acids and bases relative strengths of acids and bases, the Lux-flood, solvent system and Lewis concepts of acids and bases.** **HSAB concept:** Classification of Acids and Bases According to HSAB Theory (Hard, Borderline, Soft). Applications of HSAB Theory in Inorganic Reactions – Solubility, Selectivity, Redox Reaction. Non-aqueous solvents: Physical properties of a solvent, types of solvents and their general characteristics, Liquid ammonia as a solvent. Acid-base, precipitation and complex, formation reactions. Solutions of alkali and alkaline earth metals in ammonia-application)	12
II	**CHEMISTRY OF C-C σ-BONDING** Alkanes: Preparation (Wurtz reaction, reduction/hydrogenation of alkenes, Corey-House method). Reactions (mechanisms): halogenation, free radical substitution. Cycloalkanes: Preparation (Dieckmanns ring closure, reduction of aromatic hydrocarbons), Reactions (mechanisms): substitution and ring-opening reactions. tability of cycloalkanes – Baeyer's strain theory, Sachse and Mohr predictions,	11

	Conformational structures of ethane, n-butane and cyclohexane. **'HEMISTRY OF C-C π- BONDING** .lkenes: Preparation methods (dehydration, dehydrohalogenation, dehydrogenation, Holffmann and Saytzeff rules, cis and trans eliminations). Reactions (mechanisms): electrophilic and free radical addition (hydrogen, halogen, hydrogen halide, hydrogen bromide, water, hydroboration, ozonolysis, dihydroxylation with $KMnO_4$).)ienes: 1,2 and 1,4-additions, Diels-Alder reactions. .lkynes: Preparation (dehdrohalogenation, dehydrogenation), Reactions: Acidity, formation of acetylides, addition of water, hydrogen halides and halogens, oxidation,)zonolysis, hydroboration/oxidation. **.romatic Hydrocarbons** .romatic hydrocarbons: Aromaticity: Huckel's rule, aromatic character of arenes, cyclic carbocations/ carnanions and heterocyclic compounds with suitable examples. Electrophilic aromatic substitution: halogenation, nitration, sulphonation and Friedel-Craft's alkylation/acylation with their mechanism. Directive effects of the groups.	
III	**Behaviour of ideal gases:** Kinetic theory of gases – postulates and derivation of the equation, $PV=1/3\ mne^2$ and derivation of the gas laws-Maxwell's distribution of molecular velocatities-effects of temperature-types of molecular velocities-degrees of freedom-Principle of equipartition of energy. **Behaviour of Real gases:** Deviation from ideal behaviour, derivation of van der Waals, equation of state and critical constants. Liquid state chemistry: structure of liquids (Eyring Theory), Properties of liquids, viscosity and surface tension. **Solid state chemistry:** Nature of the solid state, law of constancy of interfacial angles, law of rational indices, Miller indices, elementary ideas of symmetry, symmetry elements and symmetry operations, seven crystal systems and fourteen Bravais lattices; X-ray diffraction, Bragg's law, Crystal defects.	11
IV	**A. Colloids and surface chemistry:** Classification, Optical, Kinetic and Electrical Properties of colloids, Coagulation, HrdySchuize law, flocculation value, Protection, Gold number, Emulsion, micelles and types, Gel, Syneresis and thixotropy, Physical adsorption, chemisorption, **B.** Chemical kinetics: Rate of reaction, Factors influencing rate of reaction rate law, rate constant, Order and molecularity of reactions, rate determining step, Zero, First and Second order reactions, Rate and Rate Law, methods of determining order of reaction, Chain reactions. Temperature dependence of Reaction rate, Arrhenius theory, Physical	11

	significance of Activation energy, collision theory, demerits of collision theory, non-mathematical concept of transition state theory. Catalysis: Homogeneous and Heterogeneous Catalysis, types of catalyst, characteristics of catalyst, Enzyme catalysed reactions, Industrial applications of catalysis.	
Keywords	***Acid & bases, Alkanes, Cycloalkanes, alkenes, Dienes, alkynes, Aromatic Hydrocarbons, Kinetic theory of gases, Real gases, Intermolecular forces, Crystal structure, Chemical kinetics.***	
Signature of Convener & Members (CBoS)		

SHRI DAVARA UNIVERSITY NAYA RAIPUR

FOUR YEAR UNDERGRADUATE PROGRAM (2024-28)

DEPARTMENT OF CHEMISTRY

COURSE CURRICULUM

PART-C: Learning Resources

Text Books, Reference Books and Others

- Paula, B. Y. (2014). Organic Chemistry (7th Ed.). Pearson Education, Inc. (Singapore).
- Solomons, T. W. G. (2017). Organic Chemistry (Global Ed.). John Wiley & Sons.
- Morrison, R. T., & Boyd, R. N. (2010). Organic Chemistry (7th Ed.). Prentice-Hall Of India Limited.
- Laidler, K. J., & Meiser, J. H. (2006). Physical Chemistry (2nd Indian Ed.). CBS Publishers.
- . Atkins, P. W., & De Paula, J. (2006). Physical Chemistry (8th Ed.). Oxford University Press.
- . Dogra, S., & Dogra, S. (2006). Physical Chemistry through Problems (2nd Ed.). New Age International.

Sangaranarayanan, M. V., & Mahadevan, V. (2011). Textbook of Physical Chemistry. University Press.

Text Books Recommended-

- Bahl, A., & Bahl, B. S. (2014). Organic Chemistry (22nd Ed.). S. Chand & Sons.
- Ahhuwalia, V. K., & Goyal, M. (2001). A Textbook of Organic Chemistry. Narosa Publishing House.
- . Jain, M. K., & Sharma, S. C. (2017). Modern Organic Chemistry. Vishal Publishing Company. Puri, B. R., Sharma, L. R., & Pathania, M. S. (2013). Principles of Physical Chemistry (46th Ed.).
- Shoban Lal Nagin Chand And Co. 5. Bahl, B. S. A., & Tuli, G. D. (2009). Essentials of Physical Chemistry (Multicolour Ed.). S. Chand & Company Pvt Ltd. 6. Puri, B. R., Sharma, L. R., & Kalia, K. C. (2018). Principles of Inorganic Chemistry. Nagin Chand and Co., New Delhi,

Online Resources-

- e-books and e-learning portals
- https://bit.lv/3AvV3mZ
- https://bit.ly/30V85z
- https://bit.ly/3C9PXPS
- https://bit.ly/301p9rZ
- https://bit.ly/BPnwqe

Online Resources-
e-sources/e-books and e-learning portals

ART -D: Assessment and Evaluation

SHRI DAVARA UNIVERSITY NAYA RAIPUR

Suggested Continuous Evaluation Methods: Maximum Marks: 100 Marks Continuous Internal Assessment (CIA): 30 Marks End Semester Exam (ESE): 70 Marks		
Continuous Internal Assessment (CIA): 30 (By Course Teacher)	Internal Test/Quiz:20+20 Assignment/ Semenar-10 Total Marks-30	Better marks out of the two Tot Quiz + obtained marks in Assignment shall be considered against 15 Marks
End Semester Exam (ESE):70	Two section A&B Section A :Q1 Objective 10∗1=10 Marks Q2 Short answer type-5*4=20 Section B : Descriptive answer type qts 1 out of 2frm each- 4*10=40 Marks	
Signature of Convener & Members (CBoS)		

SHRI DAVARA UNIVERSITY NAYA RAIPUR

FOUR YEAR UNDERGRADUATE PROGRAM (2024-28)

DEPARTMENT OF CHEMISTRY

COURSE CURRICULUM

PART- A: Introduction			
Program: Bachelor in Sciences **(Certificate/Diploma/Degree/Honors**		**Semester-II**	**Session: 2024-2025**
Course Code	**CHSC-02P**		
Course Title	**Lab. Course -01 (FUNDAMENTAL CHEMISTRYII)**		
Course Type	**Laboratory course**		
Pre-requisite(if any)	**As per program**		
Course Learning. Outcomes (CLO)	**At the end of this course, the students will be able to** ➢ Demonstrating and using common glassware for accurate measurements. ➢ Studying the functional group analysis organic compounds. ➢ points to assess compound purity and employing distillation and sublimation techniques to establish boiling points. ➢ Equipping with essential skills in measuring liquid surface tension and solution viscosity.		
Credits Value	**1 Credits**	**Credit =30 Hours Laboratory or Field learning/Training**	
Total Marks	**Max. Marks:50**	**Min Passing Marks: 20**	
PART-B:	Content of the Course		
Total No. of learning-Training/performance Periods: 30 Periods (30 Hours)			
Module	Topics (Course contents)		No. of Period
Lab/ field Training/ Experiment Content of Course	**Basic Laboratory Techniques:** Demonstration 80-82 ˚of Laboratory Glassware and Equipment , Calibration of Thermometer : (Naphthalene), 113.5 ˚- 114 ˚C (Acetanilide), 132.5 ˚C – 133 ˚C (Urea), 100 ˚C (Distilled Water) **Functional group analysis of Organic Compounds,** Detection of elements (N,S and halogens) and Functional groups. **Physical Chemistry** Surface tension measurements: Determine the surface tensionby (i) drop number (ii) drop weight method. Surface tension composition curve for a binary liquid mixture. Viscosity measurement using Ostwald's viscometer, Determination of viscosity of aqueous solutions of (i) sugar (ii) ethanol at room temperature. Study of the variation of viscosity of surface solution with th concentration of solute. Viscosity Composition curve for a binary liquid mixture.		30
Keywords	**Basic laboratory techniques ,Equipment's, Calibration, Melting points, Qualitative analysis, physical chemistry, Surface tension, Viscosity.**		
Signature of Convener & Members (CBoS)			

SHRI DAVARA UNIVERSITY NAYA RAIPUR

FOUR YEAR UNDERGRADUATE PROGRAM (2024-28)

DEPARTMENT OF CHEMISTRY

COURSE CURRICULUM

PART-C: Learning Resources

Text Books, Reference Books and Others

Text Books Recommended-

1. Gurtu, J. N., & Kapoor, R. (1987). Experimental Chemistry. S. Chand & Co
2. Bajpai, D. N., Pandey, O. P., & Giri, S. (2013). Practical Chemistry. S. Chand & Co.
3. 3. Ahluwalia, V. K., Dhingra, S., & Dhingram, S. (2005). College Practical Chemistry. Universities.
4. Kamboj, P. C. (2014). Advanced University Practical Chemistry (Part 1). Vishal Publishing Co. 5. Fultariya, C., & Harsora, J. (2017). Volumetric Analysis: Conceptt and Experiments.

Reference Books Recommended-

1. Mcpherson, P. A. (2015). Practical Volumetric Analysis. Royal Society Of Chemistry.
2. Shobha, R., & Banani, M. (2017). Essentials of Analytical Chemistry. Pearson.
3. Venkateswaran, V., Veeraswamy, R., & Kulandaivelu, A. R. (2004). Basic Principles Of Practical Chemistry (2nd Ed.). S. Chand Publications.
4. Sundaram, S., & Raghavan, K. (1996), Practical Chemistry. S. Viswanathan Co. Pvt. 5. Svehla, G. (2011). Vogel's Textbook of Inorganic Qualitative Analysis (7th Ed.). Pearson Education

Online Resources-

- E-resources/e-books and e-learning portals
- http://www.swayam.ac.in
- http://www.ignou.ac.in
- www.egyankosh.ac.in
- www.litm.ac.in
- www.eskillindia.org
- www.eshiksha.mp.gov.in
- www.vlab.co.in

Online Resources-
e-sources/e-books and e-learning portals

- https://www.ncbi.nlm.nih.gov/pmc/articles/PMC5871155/
- https://cms.botany.org/home/careers-jobs/careers-in-botany/arcas-of-specialization- in-botany.html

PART -D: Assessment and Evaluation

Suggested Continuous Evaluation Methods:
Maximum Marks: 50 Marks
Continuous Internal Assessment (CIA): 15 Marks
End Semester Exam (ESE): 35 Marks

Continuous Internal Assessment (CIA): 15 (By Course Teacher)	Internal Test/Quiz:10+10 Assignment/ Semenar-05 Total Marks-15	Better marks out of the two Tot Quiz + obtained marks in Assignment shall be considered against 15 Marks

End Semester Exam (ESE):35	Laboratory/Field Skill Performance: On spot Assessment Section A : Performed the Task based on lab, work 20∗1=20 Marks B: Performed the Task based on lab, work (written) 10*1=10Marks Section B : Viva-voce (based on principle/technology) - 5*1=05 Marks

Signature of Convener & Members (CBoS)

SHRI DAVARA UNIVERSITY NAYA RAIPUR

FOUR YEAR UNDERGRADUATE PROGRAM (2024-28)

DEPARTMENT OF PHYSICS

COURSE CURRICULUM

PART-A: Introduction			
Program: Bachelor in Sciences (Certificate/Diploma/Degree Honors)		**Semester-II**	**Session: 2024-2025**
Course Code	**PHSC-2T**		
Course Title	**Electricity and Magnetism**		
Course Type	**Discipline Specific course (DSC)**		
Pre-requisite(if any)	**As per program**		
Course Learning. Outcomes (CLO)	**At the end of this course, the students will be able to** ➢ State various laws related with electrostatics, dielectric, electric current, magnetism and electromagnetic induction ➢ Apply vector (electric fields, Coulomb's law) and scalar (electric Potential, electric potential energy) formalisms of electrostatics. ➢ Compare rise and decay of current in LR, CR, LCR circuits.		
Credit Value	**3 Credits**	**Credit =15 Hours-learning & Observation**	
Total Marks	**Max. Marks:=100**	**Min Passing Marks: 40**	
PART -B: Content of the Course			
Total No. of Teaching-learning Periods (01 Hr. per period) -45 Periods (45 Hours)			

Unit	Topics (Course contents)	
I	**Power plants in Chhattisgarh:-** An overview of thermal and hydroelectric power plants in Chhattisgarh. **Vector Analysis:-** Divergence & Curl of Vector fields, Line, surface and volume integrals of Vector fields, Gauss-divergence theorem and Stoke's theorem of vectors and its application in electrostatics and magneto statics. **Electrostatics field:-** Electrostatic Field, electric flux, Gauss's theorem of electrostatics, Applications of Gauss theorem- Electric field due to point charge, infinite line of charge, plane charged sheet, charged conductor.	12
II	**Electrostatic potential:-** Electric potential as line integral of electric field, potential due to a point charge, Calculation of electric field from potential, Capacitance of Parallel plate capacitor, Energy per unit volume in electrostatic field. **Dielectric & Electric Currents:-** Dielectric medium, Polarisation, Displacement vector, Gauss's theorem in dielectrics, Parallel plate capacitor completely filled with. dielectric. Steady current,	11

	current density J, non-steady current and Continuity equation, Rise and decay of current in LR, CR, LCR circuits.	
III	**Magnetism:-** Biot-Savart's law and its applications- straight conductor, circular coil, solenoid carrying current, Divergence and curl of magnetic field, Magnetic vector potential, Ampere's circuital law,Magnetic properties of materials: Magnetic intensity, magnetic induction, permeability, magnetic susceptibility, Brief introduction of dia, para and ferro-magnetic materials.	11
IV	**Electromagnetic Induction:-** Faraday's laws of electromagnetic induction, Lenz's law, self and mutual inductance, L of single coil, M of two coils, Energy stored in magnetic field. **Maxwell's equations and Electromagnetic wave propagation:-** Equation of continuity of current, Displacement current, Maxwell's equations, Wave equation in free space.	11
Keywords	**Vector calculus, Electrostatics, Dielectrics and Electric Current, Magnetism, Electromagnetic Induction, Maxwell's Equation and Electromagnetic Wave Propagation.**	
Signature of Convener & Members (CBoS)		

SHRI DAVARA UNIVERSITY NAYA RAIPUR

FOUR YEAR UNDERGRADUATE PROGRAM (2024-28)

DEPARTMENT OF PHYSICS

COURSE CURRICULUM

PART-C: Learning Resources

Text Books, Reference Books and Others

1. Electricity and Magnetism, D C Tayal, 1988, Himalaya Publishing House.
2. Unified Physics Part II, R. P.Goyal, Shivlal Agrawal and Sons
3. Unified Physics-Navbodh Publications
4. Introduction to Electrodynamics and Electromagnetism, H.C.Verma,

Text Books Recommended-

1. Vector analysis-Schaum's Outline, M.R. Spiegel, S. Lipschutz, D. Spellman, 2nd Edn., 2009, McGraw-Hill Education.
2. University Physics, Ronald Lane Reese, 2003, Thomson Brooks/Cole.

Online Resources-

- e-books and e-learning portals
- http://www.swayam.ac.in
- http://www.ignou.ac.in
- http://www.egvankosh.ac.in
- http://www.itm.sc.in
- http://www.eskillindia.org
- http://www.eshiksha.mp.gov.in
- http://www.viah.co.in
- http://www.internshala.com

Online Resources-
e-sources/e-books and e-learning portals

- https://www.pbs.org/video/botany-basics-iuu2bl/
- https://efaidohmannibpcapcalclefindorkaj/https://www2.ca.uky.edu/apcom/pubs/ho/ho96/ho96.pdf
- https://www.botanytoday.com/brunches-of-botany

PART -D: Assessment and Evaluation

Suggested Continuous Evaluation Methods:
Maximum Marks: 100 Marks
Continuous Internal Assessment (CIA): 30 Marks
End Semester Exam (ESE): 70 Marks

Continuous Internal Assessment (CIA): 30 (By Course Teacher)	Internal Test/Quiz:20+20 Assignment/ Semenar-10 Total Marks-30	Better marks out of the two Tot Quiz + obtained marks in Assignment shall be considered against 15 Marks

SHRI DAVARA UNIVERSITY NAYA RAIPUR

End Semester Exam (ESE):70	Two section A&B Section A :Q1 Objective 10∗1=10 Marks, Q2 Short answer type-5∗4=20 Section B : Descriptive answer type qts 1 out of 2frm each- 4*10=40 Marks
Signature of Convener & Members (CBoS)	

SHRI DAVARA UNIVERSITY NAYA RAIPUR

FOUR YEAR UNDERGRADUATE PROGRAM (2024-28)

DEPARTMENT OF PHYSICS

COURSE CURRICULUM

PART- A: Introduction			
Program: Bachelor in Sciences **(Certificate/Diploma/Degree/Honors**		**Semester-II**	**Session: 2024-2025**
Course Code	**PHSC-02P**		
Course Title	**Lab.Course-02 (Electricity and Magnetism)**		
Course Type	**Laboratory course**		
Pre-requisite(if any)	**As per program**		
Course Learning. Outcomes (CLO)	**At the end of this course, the students will be able to** ➢ Assemble required parts/devices and arrange them to perform experiments. ➢ Record/ observe data as required by the experimental objectives. ➢ Analyze recorded data and formulate it to get desired results. ➢ Interpret results and check for attainment of proposed objectives related to laws of mechanics and its applications.		
Credits Value	**1 Credits**	**Credit =30 Hours Laboratory or Field learning/Training**	
Total Marks	**Max. Marks:50**	**Min Passing Marks: 20**	

PART-B:	**Content of the Course**	
Total No. of learning-Training/performance Periods: 30 Periods (30 Hours)		
Module	Topics (Course contents)	No. of Period
Lab/ field Training/ Experiment Content of Course	1. To use a Multimeter for measuring (a) Resistances, (b) AC and DC Voltages, (c) DC Current, and (d) checking electrical fuses. 2. To compare capacitances using De'Sauty's bridge. 3. Measurement of field strength B and its variation in a Solenoid Determine (dB/dx). 4 . To study the Characteristics of a Series RC Circuit. 5. To study a series LCR circuit and determine its (a) Resonant Frequency, (b) Quality Factor. 6. To study a parallel LCR circuit and determine its (a) Anti-resonant frequency and(b) Quality factor Q. 7. To determine a Low Resistance by Carey Foster's Bridge. 8. To verify the Thevenin and Norton theorem. 9. To verify the Superposition, and Maximum Power Transfer Theorem. 10. Study of magnetic field due to a current loop.	30
Keywords	**Multimeter, Capacitance Comparison, Magnetic Field, RC Circuit, Series LCR Circuit, Parallel LCR Circuit, Low Resistance Measurement, Electrical Theorems**	
Signature of Convener & Members (CBoS)		

SHRI DAVARA UNIVERSITY NAYA RAIPUR

FOUR YEAR UNDERGRADUATE PROGRAM (2024-28)

DEPARTMENT OF PHYSICS

COURSE CURRICULUM

PART-C: Learning Resources
1. Engineering Practical Physics, S.Panigrahi&B.Mallick, 2015, Cengage Learning India Pvt. Ltd. 2. A Text Book of Practical Physics, Indu Prakash and Ramakrishna, 11th Edition, 2011, Kitab Mahal, New Delhi. 3. Unified Practical Physics: RP Goyal, Shivlal Agrawal & Sons 4. Unified Practical Physics: Yugbodh Prakashan 5. Unified Practical Physics: Navbodh Prakashan
Text Books Recommended-
Reference Books Recommended- 1. Charak Samhita 1 2. "Medicinal Plants of India" by C.P. Khare
Online Resources- ➢ E-resources/e-books and e-learning portals ➢ http://www.swayam.ac.in ➢ http://www.ignou.ac.in ➢ www.egyankosh.ac.in ➢ www.litm.ac.in ➢ www.eskillindia.org ➢ www.eshiksha.mp.gov.in ➢ www.vlab.co.in
Online Resources- e-sources/e-books and e-learning portals ➢ https://www.ncbi.nlm.nih.gov/pmc/articles/PMC5871155/ ➢ https://cms.botany.org/home/careers-jobs/careers-in-botany/arcas-of-specialization- in-botany.html
PART -D: Assessment and Evaluation

SHRI DAVARA UNIVERSITY NAYA RAIPUR

Suggested Continuous Evaluation Methods: Maximum Marks: 50 Marks Continuous Internal Assessment (CIA): 15 Marks End Semester Exam (ESE): 35 Marks		
Continuous Internal Assessment (CIA): 15 (By Course Teacher)	Internal Test/Quiz:10+10 Assignment/ Semenar-05 Total Marks-15	Better marks out of the two Tot Quiz + obtained marks in Assignment shall be considered against 15 Marks
End Semester Exam (ESE):35	Laboratory/Field Skill Performance: On spot Assessment Section A : Performed the Task based on lab, work 20∗1=20 Marks B: Sporting based on lab, work (written) 10*1=10Marks Section B : Viva-voce (based on principle/technology) - 5*1=05 Marks	
Signature of Convener & Members (CBoS)		

SHRI DAVARA UNIVERSITY NAYA RAIPUR

FOUR YEAR UNDERGRADUATE PROGRAM (2024-28)

DEPARTMENT OF ZOOLOGY

COURSE CURRICULUM

PART-A: Introduction			
Program: Bachelor in Sciences (Certificate/Diploma/Degree Honors)		**Semester-II**	**Session: 2024-2025**
Course Code	**ZOSC-02T**		
Course Title	**Cell Biology and Histology**		
Course Type	**Discipline Specific course (DSC)**		
Pre-requisite(if any)	**As per program**		
Course Learning. Outcomes (CLO)	**After successfully completing this course, the students will be able to –** ➢ **Acquire knowledge of Cell membrane and function** ➢ **Understand the functioning of nucleus and extra nuclear organelles and understand the intricate cellular mechanisms involved.** ➢ **Gain Knowledge of key processes like cell division,** ➢ **Learn about various tissues of body their structural significance**		
Credit Value	**3 Credits**	**Credit =15 Hours-learning & Observation**	
Total Marks	**Max. Marks:=100**	**Min Passing Marks: 40**	

PART -B: Content of the Course		
Total No. of Teaching-learning Periods (01 Hr. per period) -45 Periods (45 Hours)		
Unit	Topics (Course contents)	
I	**Cell Structure, Cell Membrane and Extra Nuclear Cell Organelles:** General structure of Prokaryotes and Eukaryotes. Cell membrane organization: Origin, structure (Lipid-Lipid Bilayer Model, Dannelli & Davson Model, Unit Membrane Model and Fluid mosaic model), chemical composition and function of cell membrane, Specialization of cell membrane: microvilli desmosomes, Hemidesmosome, Septate Desmosome, plasmodesmata, tight and gap junction. **Extra Nuclear Cell Organelles:** Ultra structure and functions of Endoplasmic reticulum and Golgi apparatus.	12
II	**Extra Nuclear Cell Organelles:** Ultra structure and functions of Ribosome, Lysosome, Peroxisomes, Mitochondria: Origin, structure and function.	11
III	**Nuclear Organization and Cell Division:** Size, shape, structure and functions of interphase nucleus. Ultra structure of nuclear membrane and pore complex. Nucleolus: general organization, chemical composition and functions, Chromosome Morphology, Cell cycle, Cell division- Mitosis and Meiosis. Cell division checks points and their regulation. Programmed cell death (Apoptosis).	11
IV	**Introduction to tissues. Epithelial tissue:** types, structure and characteristics. surface modifications. Basement membrane: structure and characteristics. Connective tissue cells. Structure and function of loose, dense and adipose tissue.	11

	Cartilage and bone: classification, and fine structure. Blood: plasma, blood cells, lymph- their structure and function. Bone marrow and hemopoiesis. Structure and function of spleen. Muscular tissue: ultrastructure of smooth, skeletal and cardiac muscles. Muscle-tendon attachment. Structure and classification of neurons.	
Keywords	**Cell Biology, Cell Membrane, Cell organelle, Nucleus, endoplasmic reticulum and Golgi apparatus, ribosome, lysosome, peroxisomes, Mitochondria, tissues.**	
Signature of Convener & Members (CBoS)		

SHRI DAVARA UNIVERSITY NAYA RAIPUR

FOUR YEAR UNDERGRADUATE PROGRAM (2024-28)

DEPARTMENT OF ZOOLOGY

COURSE CURRICULUM

PART-C: Learning Resources

Text Books, Reference Books and Others

Text Books Recommended-

1. E. J. W. Barrington, Invertebrate structure and function, English Language Book Society UK.
2. Robert Barnes, Invertebrate Zoology, Robert Barnes IVth edition Holt Saunders International Edition Japan.
3. Park Haswell, Marshall and Williams, A textbook on Zoology Invertebrate, AITBS Publishing and Distributers, Delhi.
4. Park Haswell, Marshall and Williams, A textbook on Zoology Vertebrate, ATTBS Publishing and Distributers, Delhi.

Reference Books Recommended-

1. ProfR. L. Kotpal, Protozoa to Echinodermata, Rastogi Publication Meerut.
2. EL. Jordan, Dr. P. S. Verma, Invertebrate Zoology, S. Chand Publications, New Delhi.
3. N. Arumugam, N. C. Nair S. - Invertebrate Zoology, Saras Publication..
4. Barrington E. J. W., Invertebrate Structure and Function, Nelson London.
5. Barnes, R. D., Invertebrate Zoology-Saunders Philadelphia.
6. R. L. Kotpal, Invertebrate, Rastogi Publications R. I.. Kotpal, Vertebrate, Rastogi Publications.
7. H. S. Bhampah, KavitaJuneja, Recent trends in vertebrates vol 1-9, Anmol Publication.
8. S. N. Prasad, Life of invertebrates, Vikash Publication House Pvt Ltd New Delhi.
9. G. S. Sandhu, Harshwardhan Bhagskar-Advanced invertebrate zoology-Campus books international.

Online Resources-

- e-books and e-learning portals
- https://www.coursera.org/lecture/emergence-of-life/4-5-invertebrates-successes-of-life-http://www.ignou.ac.in
- https://www.shiksha.com/online-courses/introduction-to-biology-biodiversity-course-http://www.itm.sc.in
- https://www.youtube.com/watch?v=uK-XYhttp://www.eshiksha.mp.gov.in
- https://www.youtube.com/watch?v=WxMSckEcio4http://www.internshala.com

Online Resources-
e-sources/e-books and e-learning portals

- https://www.pbs.org/video/botany-basics-iuu2bl/
- https://efaidohmannibpcapcalclefindorkaj/https://www2.ca.uky.edu/apcom/pubs/ho/ho96/ho96.pdf
- https://www.botanytoday.com/brunches-of-botany

PART -D: Assessment and Evaluation

Suggested Continuous Evaluation Methods:

Maximum Marks:	100 Marks
Continuous Internal Assessment (CIA):	30 Marks
End Semester Exam (ESE):	70 Marks

Continuous Internal Assessment (CIA): 30 (By Course Teacher)	Internal Test/Quiz:20+20 Assignment/ Semenar-10 Total Marks-30	Better marks out of the two Tot Quiz + obtained marks in Assignment shall be considered against 15 Marks

End Semester Exam (ESE):70	Two section A&B Section A :Q1 Objective 10*1=10 Marks Q2 Short answer type-5*4=20 Section B : Descriptive answer type qts 1 out of 2frm each- 4*10=40 Marks

Signature of Convener & Members (CBoS)

SHRI DAVARA UNIVERSITY NAYA RAIPUR

FOUR YEAR UNDERGRADUATE PROGRAM (2024-28)

DEPARTMENT OF ZOOLOGY

COURSE CURRICULUM

PART- A: Introduction			
Program: Bachelor in Sciences **(Certificate/Diploma/Degree/Honors**		**Semester-II**	**Session: 2024-2025**
Course Code	**BOSC-02P**		
Course Title	**Lab. Course -03 Cell Biology and Histology**		
Course Type	**Laboratory course**		
Pre-requisite(if any)	**As per program**		
Course Learning. Outcomes (CLO)	**After successfully completing this course, the students will be able to -** ➢ **Understand ultra structure of prokaryote and Eukaryote cell, undertake microscopic study to gain knowledge** ➢ **learn to identify cell organelles** ➢ **Explain and demonstrate mitosis and meiosis division in onion root tip, Grass hopper testis, etc** ➢ **Gain knowledge of Microtomy**		
Credits Value	**1 Credits**	**Credit =30 Hours Laboratory or Field learning/Training**	
Total Marks	**Max. Marks:50**	**Min Passing Marks: 20**	
PART-B:	**Content of the Course**		
Total No. of learning-Training/performance Periods: 30 Periods (30 Hours)			

Module	Topics (Course contents)	No. of Period
Lab/ field Training/ Experiment Content of Course	1.Study of prokaryotic and eukaryotic cell types with the help of chart, slide and video. 2.Separation and isolation of cells by sedimentation velocity in unit gravity. 3. Disruption of cells, isolation and identification of subcellular components, isolation of nuclei. 4. Isolation of mitochondria by differential centrifugation and identification of succinic dehydrogenase in the mitochondrial pellet. 5. Chromosome segregation in mitosis and meiosis. 6. Preparation of chromosome squashes from Onion Root tip for observation of stages of Mitosis 7. Preparation of chromosome squashes from grasshopper/cockroach testes for the observation of stages of meiosis. 8. Isolation and estimation of DNA.	30

Commented [u1]:

	9. Study of types of tissue through permanent slides: epithelial, connective, muscular, Nervous etc. 10. Preparation of Practical Record 11. Group discussion/Viva or Seminar presentation on related topics mentioned in Theory paper	
Keywords	**Prokaryote, Eukaryote, cell division, Mitosis, Meiosis, DNA Separation, Histology of Tissue, Microtomy**	
Signature of Convener & Members (CBoS)		

SHRI DAVARA UNIVERSITY NAYA RAIPUR

FOUR YEAR UNDERGRADUATE PROGRAM (2024-28)

DEPARTMENT OF ZOOLOGY

COURSE CURRICULUM

PART-C: Learning Resources

Text Books, Reference Books and Others

Text Books Recommended-

1. S.S. Lal, Practical Zoology, Invertebrate. 12 Edition Rastogi Publications, Meerut, New Delhi.
2. A manual of practical Zoology. Dr. P.S Verma, S. Reference Books Recommended- Chand Publication, New Delhi.

Reference Books Recommended-

1. Park Haswell, Marshall and Williams, A textbook on Zoology Invertebrate, AITBS Publishing and Distributers, Delhi .
2. Park Haswell, Marshall and Williams, A textbook on Zoology Vertebrate, AFTBS Publishing and Distributers, Delhi.

Online Resources-

- E-resources/e-books and e-learning portals
- http://ndi.ätkgp.ac.in/he/document/swayamprabha/swayam
- http://www.swayam.ac.in
- http://www.ignou.ac.in
- www.egyankosh.ac.in
- www.litm.ac.in
- www.eskillindia.org
- www.eshiksha.mp.gov.in

Online Resources-

e-sources/e-books and e-learning portals

- https://www.ncbi.nlm.nih.gov/pmc/articles/PMC5871155/
- https://cms.botany.org/home/careers-jobs/careers-in-botany/arcas-of-specialization- in-botany.html

PART -D: Assessment and Evaluation

Suggested Continuous Evaluation Methods:
Maximum Marks: 50 Marks
Continuous Internal Assessment (CIA): 15 Marks
End Semester Exam (ESE): 35 Marks

Continuous Internal Assessment (CIA): 15 (By Course Teacher)	Internal Test/Quiz:10+10 Assignment/ Semenar-05 Total Marks-15	Better marks out of the two Tot Quiz + obtained marks in Assignment shall be considered against 15 Marks

End Semester Exam (ESE):35	Laboratory/Field Skill Performance: On spot Assessment Section A : Performed the Task based on lab, work 20∗1=20 Marks B: Spotting frased on tools & technology (written) 10*1=10Marks Section B : Viva-voce (based on principle/technology) - 5*1=05 Marks

Signature of Convener & Members (CBoS)

SHRI DAVARA UNIVERSITY NAYA RAIPUR

FOUR YEAR UNDERGRADUATE PROGRAM (2024-28)

DEPARTMENT OF POLITICAL SCIENCES

COURSE CURRICULUM

PART-A: Introduction			
Program: Bachelor in Sciences (Certificate/Diploma/Degree Honors)		**Semester-II**	**Session: 2024-2025**
Course Code	**PSGE-02**		
Course Title	**Constitutional Government in India**		
Course Type	**Discipline General Elective course (GE)**		
Pre-requisite(if any)	**As per program**		
Course Learning. Outcomes (CLO)	**After completion of the course, the student shall be able to..** ➢ **Construct the political ideals mentioned in the preamble of the constitution.** ➢ **Assess the provisions of citizenship, fundamental rights and duties and their correlation.** ➢ **Examine the role of president and the functioning of union executive.** ➢ **Interpret the provisions and functioning of the union legislature and constitutional bodies of functional democracy, like election commission, finance commission and C&AG.**		
Credit Value	**4 Credits**	**Credit =60 Hours-learning & Observation**	
Total Marks	**Max. Marks:=100**	**Min Passing Marks: 40**	

PART -B: Content of the Course		
Total No. of Teaching-learning Periods (01 Hr. per period) -45 Periods (45 Hours)		
Unit	Topics (Course contents)	
I	**Constitution Citizenship and Rights** Making of Indian Constitution: Cabinet mission plan and Constituent assembly. Preamble, features, Sources. Schedules, citizenship. Fundamental Rights and Duties, Directive Principles of State Policy. Constitution Amendment Process.	12
II	**Union** President, Vice President, Council of Ministers and Prime Minister. Federal Parliament Lok Sabha and Rajya Sabha. Supreme court Organization Functions, Powers, Judicial Review.	11
III	**Union and Federal administration** controller and auditor general Centre State Relations: Legislative, Financial, Administrative. Union and state public service commission, Election Commission, Finance Commission.	11
IV	**State and Local self government** Legislature, Executive: Governor, Council of Ministers and Chief Minister. State High Court-Organization. Functions, Rights.	11

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Keywords	Political theory, sate, sovereignty, right, they, democracy, constitution, party.
Signature of Convener & Members (CBoS)	

SHRI DAVARA UNIVERSITY NAYA RAIPUR

FOUR YEAR UNDERGRADUATE PROGRAM (2024-28)

DEPARTMENT OF POLITICAL SCIENCE

COURSE CURRICULUM

PART-C: Learning Resources

Text Books, Reference Books and Others

Text Books Recommended-

1. Ambadatt Pant Harimohan Jain Madan Gopal (1985) Fundamentals of Political Science, Central Publishing House Allahabad. U.P.
2. Sandhu Man Singh (1956) Political Theory Hindi Medium Implementation Directorate, Delhi University, New Delhi
3. Johari JC 1916) Basic principles of political science, Sahitya Bhavan, Agra.
4. Rajeev and Ashok Acharya (Eds) Political Theory A Flag, Dilsey Pearson, 2008

Reference Books Recommended-

1 umar, Sanjeev (Ed. Understanding of Political Theory, Delhi: Orient Book Swan, 2019
2 Hussain Shakeel (2018) Conceptual Introduction to Political Theory. Chhattisgarh State Hindi Forest Academy, Rampur.
3 K.K. Mishra (2010) Political Theory, 5. Chand Publishing Delhi
4 OP Gouba (2014) An Introduction to Political Theory, MacMillan Publishers, Delhi

Online Resources-

- e-books and e-learning portals
- https://www.coursera.org/lecture/emergence-of-life/-http://www.ignou.ac.in
- https://www.shiksha.com/online-courses/-http://www.itm.sc.in
- https://www.youtube.com/watch?v=uK-XYhttp://www.eshiksha.mp.gov.in
- https://www.youtube.com/watch?v=WxMSckEcio4http://www.internshala.com

Online Resources-
e-sources/e-books and e-learning portals

- https://www.pbs.org/video/political-basics-iuu2bl/
- https://efaidohmannibpcapcalclefindorkaj/https://www2.ca.uky.edu/apcom/pubs/ho/ho96/ho96.pdf
- https://www.botanytoday.com/brunches-of-botany

RT -D: Assessment and Evaluation

Suggested Continuous Evaluation Methods:

Maximum Marks:	100 Marks
Continuous Internal Assessment (CIA):	30 Marks
End Semester Exam (ESE):	70 Marks

Continuous Internal Assessment (CIA): 30 (By Course Teacher)	Internal Test/Quiz:20+20 Assignment/ Semenar-10 Total Marks-30	Better marks out of the two Tot Quiz + obtained marks in Assignment shall be considered against 15 Marks

End Semester Exam (ESE):70	Two section A&B Section A :Q1 Objective 10*1=10 Marks Q2 Short answer type-5*4=20 Section B : Descriptive answer type qts 1 out of 2frm each- 4*10=40 Marks

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SHRI DAVARA UNIVERSITY NAYA RAIPUR

FOUR YEAR UNDERGRADUATE PROGRAM (2024-28)

DEPARTMENT OF HINDI

COURSE CURRICULUM

PART-A: Introduction			
Program: Bachelor in Sciences (Certificate/Diploma/Degree Honors)		**Semester-II**	**Session: 2024-2025**
Course Code	**AEC-02**		
Course Title			
Course Type	**Ability Enhancement Course**		
Pre-requisite(if any)	**As per program**		
Course Learning. Outcomes (CLO)	**After the completion of this course, the students will be able to-** ➢ विद्यार्थी हिन्दीभाषा एव ंव्याकरण संबंधीज्ञान से समृद्ध होंगे। ➢ भाषा ज्ञान के माध्यम से भारतीय संस्कृति एवं भावनात्मक एकता के महत्व को समझने की क्षमता विकसित हो सकेगी। ➢ मुहावरे एव ंलोकोक्तियाँ का महत्व समझ सकेंगे। व्यंग्य, निबंध एवं कविता विद्या से परिचित होंगे। ➢ निबंध लेखन एवं अपठित गद्यांश के माध्यम से विद्यार्थिर्यो ंका बौद्धिक विकास हो सकेगा।		
Credit Value	**2 Credits**	**Credit =30 Periods -learning & Observation**	
Total Marks	**Max. Marks:=50**	**Min Passing Marks: 20**	

PART -B: Content of the Course		
Total No. of Teaching-learning Periods (45 Min. per period) -30 Periods		
Unit	Topics (Course contents)	
I	**रचनाएं** **भारत वंदना–सूर्यकांत त्रिपाठी 'निराला' (कविता)** **जीव–हरिशंकर परसाई (व्यंग्य) चोरी और प्रायश्चित–महात्मागांधी (निबंध)**	08
II	**हिन्दी व्याकरण एवं शब्द रचना** **प्रत्यय, संधि समास** **पर्यायवाची शब्द, विलोम शब्द, अनेकार्थी शब्द, समश्रुत शब्द, अनेक शब्दों के लिए एक शब्द**	07
III	**हिन्दी व्याकरण एवं रचनापक्ष** **मुहावरे एवं लोकोक्तियां** **परिभाशिक षब्दावली एवं हिन्दी में पदनाम, षब्द षुद्धि, वाक्य षुद्धि**	08
IV	**रचनात्मक लेखन** **निबंध लेखन** **अपठित गद्यांश** (नोट– विद्यार्थी को किसी एक विशय पर निबंध व प्रदत्त गद्यांष का षिर्शक तथा सारांष लिखना होगा।)	07

SHRI DAVARA UNIVERSITY NAYA RAIPUR

Keywords	रचनात्मक लेखन निबंध लेखन हिन्दी व्याकरण एवं रचना पक्ष
Signature of Convener & Members (CBoS)	

SHRI DAVARA UNIVERSITY NAYA RAIPUR

YEAR UNDERGRADUATE PROGRAM (2024-28)

DEPARTMENT OF HINDI

COURSE CURRICULUM

PART-C: Learning Resources

Text Books, Reference Books and Others

Text Books Recommended-

1- भारतीयता के अमर स्वर– डॉ. धनंजय वर्मा, मध्यप्रदेश हिन्दी अकादमी
2- आधुनिक हिन्दी व्याकरण और रचना– डॉ वासुदेव नंदन
3- हिन्दी भाषा और व्यवहार– डॉ. गंगा चरण त्रिपाठी
4- हिन्दी व्याकरण माला– डॉ. के.आर. गहिया, डॉ विमलेश शर्मा
5- हिन्दी व्याकरण– कामता प्रसाद गुरू

Online Resources-

- Applying Communication Theory for Professional Life: A Practical Introduction. Dainton and Zelley, http://taime.uz.ac.zw/claroline/backends/download.php?urf-L0ludHJvX3RvX2NvbW1lbmljYXRpb25f
- https://www.coursera.org/lecture/emergence-of-life/4-5-invertebrates-successes-of-life-http://www.ignou.ac.in
- https://web.sol.du.ac.in/my modules/type/cbcs-11- 2/data/root/B.Com/Semester%202/ABILITY-ENHANCEMENT%20COMPULSORY%20COURSE-AECC/English%20Communication%20A-B-C/Unit%201-5.
  - pdf https://www.youtube.com/watch?v=uK-XYhttp://www.eshiksha.mp.gov.in
  - https://www.youtube.com/watch?v=WxMSckEcio4http://www.internshala.com
  - https://archive.org/details/personality-development-book/mode/lup
  - https://www.coursera.org/articles/presentation-skills
  - https://www.cbs.de/en/blog/15-effective-presentation-tips-to-improve-presentation-skills/
  - https://benjaminball.com/blog/good-body-language-best-visual-aid-talks/

Online Resources-

- e-sources/e-books and e-learning portals https://blog.moderngov.com/importance-of-body-language-in-presentations-good-bad-
- https://efaidohmannibpcapcalclefindorkaj/https://www2.ca.uky.edu/apcom/pubs/ho/ho96/ho96.pdf
- https://www.botanytoday.com/brunches-of-botany

PART -D: Assessment and Evaluation

Suggested Continuous Evaluation Methods:
Maximum Marks: 50 Marks
Continuous Internal Assessment (CIA): 15 Marks
End Semester Exam (ESE): 35 Marks

Continuous Internal Assessment (CIA): 15 (By Course Teacher)	Internal Test/Quiz:10+10 Assignment/ Semenar-05 Total Marks-15	Better marks out of the two Tot Quiz + obtained marks in Assignment shall be considered against 15 Marks

End Semester Exam (ESE):70	Two section A&B Section A :Q1 Objective 1*5=5 Marks Q2 Short answer type-2*5=10 (I. Vocabulary, II Unseen passage Section B : Descriptive answer type qts 1 out of 2frm each- 5*4=20 Marks

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SHRI DAVARA UNIVERSITY NAYA RAIPUR

FOUR YEAR UNDERGRADUATE PROGRAM (2024-28)

DEPARTMENT OF CHEMISTRY

COURSE CURRICULUM

PART-A: Introduction			
Program: Bachelor in Sciences (Certificate/Diploma/Degree Honors)		**Semester-II**	**Session: 2024-2025**
Course Code	**SEC-01**		
Course Title	**Chemistry Lab Skills -II**		
Course Type	**Skill Enhancement Course (SEC)**		
Pre-requisite(if any)	**As per program**		
Course Learning. Outcomes (CLO)	**Completing this course, students will be able to: -** ➢ **To understand different acid-base theories and solvent system.** ➢ **To learn the preparation, bonding, and reactions of C-C σ- & π-bonded compounds.** ➢ **To understand the concept and chemistry of aromatic compounds and their reactions.** ➢ **To learn the basic concepts of various states of matter & understand the basic concepts of surface chemistry and chemical kinetics.**		
Credit Value	**1 Credits**	**Credit =30 Periods -learning & Observation**	
Total Marks	**Max. Marks:=50**	**Min Passing Marks: 20**	

PART -B: Content of the Course		
Total No. of Teaching-learning Periods (45 Min. per period) -15 Periods		
Unit	Topics (Course contents)	
I	**Introduction of Chemistry Laboratory** General introduction of the chemistry laboratory, common instructions for safe working in chemical laboratories, Good Laboratory Practices (GLP), Good Manufacturing Practices (GMP).Laboratory design, Storage, ventilation, lighting, fume, cupboard, arrangement of the store, Safety provisions, Organization of practical work, Maintenance of laboratory, equipment Cleaning of laboratories and glasswares / plasticwares and preparation room. Classification of apparatus in store and laboratory.	08
II	**Introduction of Chemistry Apparatus** Glass apparatus - Beaker, test tube, boiling tube, funnel, separating funnel,	07

	filtration flask, round bottom flask, flat bottom flask, condenser Liebig flask, watglass etc. measuring conical or condenser, Petridis, desiccators. Volumetric Apparatus – Measuring cylinder, burette, pipette, volumetric flask, analytical balance, single-pan electronic balance/ electrical analytical balance, Micropipette, Three way Pipette Bulb etc.	
III	**Introduction of Chemistry Equipments** Clevenger apparatus, Buchner funnel, Soxhlet extractor, wire gauze, cork borers, filter pumps, crucible, mohr clip, pipe clay triangle, pestle and mortar, sprit lamp,spatulas, thermometer, pH meter.	08
IV	**Introduction of Chemistry Equipments**- laboratory centrifuge. Apparatus for heating and reaction: Magnetic Stirrer, Bunsen burner, water bath, oil bath hot plate, sand ,bath, hot air oven, heating mantle etc.	07
Keywords	***Introduction of Chemistry Laboratory.*** **Introduction of Chemistry Apparatus. Introduction of Chemistry Equipment's.**	
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SHRI DAVARA UNIVERSITY NAYA RAIPUR

FOUR YEAR UNDERGRADUATE PROGRAM (2024-28)

DEPARTMENT OF CHEMISTRY

COURSE CURRICULUM

PART-C: Learning Resources		
Text Books, Reference Books and Others		
Text Books Recommended-		
Reference Books Recommended- ➢ Bahl, A., & Bahl, B. S. (2014). Organic Chemistry (22nd Ed.). S. Chand & Sons. ➢ Ahhuwalia, V. K., & Goyal, M. (2001). A Textbook of Organic Chemistry. Narosa Publishing House. ➢ . Jain, M. K., & Sharma, S. C. (2017). Modern Organic Chemistry. Vishal Publishing Company. Puri, B. R., Sharma, L. R., & Pathania, M. S. (2013). Principles of Physical Chemistry (46th Ed.). ➢ Shoban Lal Nagin Chand And Co. 5. Bahl, B. S. A., & Tuli, G. D. (2009). Essentials of Physical Chemistry (Multicolour Ed.). S. Chand & Company Pvt Ltd. 6. Puri, B. R., Sharma, L. R., & Kalia, K. C. (2018). Principles of Inorganic Chemistry. Nagin Chand and Co., New Delhi,		
Online Resources- ➢ Introduction to Computer Fundamental from W3school: https://www.w3schools.blog/computer-fundamentals-tutorial ➢ Introduction to MS-Word from W3school: https://www.w3schools.blog/ms-word-tutorial ➢ Introduction to MS-Excel from W3school:https://www.w3schools.com/excel/excel_introduction.php ➢ ntroduction to MS-PowerPoint from W3school:https://www.w3schools.blog/powerpoint-tutorial Introduction to MS-Access from W3school: ➢ https://www.youtube.com/watch?v=WxMSckEcio4http://www.internshala.com		
Online Resources- e-sources/e-books and e-learning portals ➢ https://www.rgycsm.org/uploads/books/MICROSOFT-OFFICE-BOOK.pdf ➢ https://www.youtube.com/watch?v=SH40YV5AJ6A ➢ https://www.youtube.com/watch?v=SH40YV5AJ6A ➢ https://hte.rajasthan.gov.in/dept/dte/board		
PART -D: Assessment and Evaluation		
Suggested Continuous Evaluation Methods: Maximum Marks: 50 Marks Continuous Internal Assessment (CIA): 15 Marks End Semester Exam (ESE): 35 Marks		
Continuous Internal Assessment (CIA): 15 (By Course Teacher)	Internal Test/Quiz:10+10 Assignment/ Semenar-10 Total Marks-15	Better marks out of the two Tot Quiz + obtained marks

		in Assignment shall be considered against 15 Marks
End Semester Exam (ESE):50	Two section A&B Section A :Q1 Objective 5∗1=5 Marks Q2 Short answer type-5∗4=20 Section B : Descriptive answer type qts 1 out of 2frm each- 10*1=10 Marks	
Signature of Convener & Members (CBoS)		